

Pharmacognosy III, 3rd stage, 2nd Semeter



<u>Quantitative and Qualitative</u> <u>Analysis of Piperine</u> <u>Alkaloid</u>

ANALYSIS OF PIPERINE ALKALOID

QUALTITATIVE ANALYSIS



hemical Test Specific Test Chromatography T.L.C QUANTITATIVE ANALYSIS ✓ By weighting the crystals



& Wagner 's Test:

Aim: To indicate in general the alkaloid as other alkaloids.

Equipments and Reagents:

- & Test tube.
- & Ethanol.
- &HCl.
- & Procedure:

1- Preperation of Wagners reagent: This reagent is prepared by dissolving 1 gm of iodine and 3 gm of potassium iodide in 50 mL of distilled water. This test is utilized to detect alkaloids, yielding a reddishbrown precipitate by reaction.

2- Take few crystals of piperine alkaloid and dissolve in few mls of ethanol, in test tube then add 2 drops of HCl. Then add 2 drops of Wagner's reagent.
 Result: Brown precipitate will occur.

A: Wagner's reagent. B. Result.





Proposed reaction for Wagner test + I^O ____ Į, I, $+ KI + I_2$ ĸ⊕ " ChemistNotes.com Reddish-Brown ppt.

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The Identification of Piperine Alkaloid By Chromatography (TLC)

& By the use of thin layer chromatography (T.L.C)
& The stationary phase = Silica gel G.
& The mobile phase =Toluene: Diethylether: Diaxon (62.5:21.5:16).
& The spray reagent =Dragendorff's reagent.
& Mechanism of separation = Adsorption.
& Developing = Ascending.

Procedure:

1- Prepare 100ml of mobile phase, and place it in the glass tank.

- 2- Cover the tank with glass lid and allow standing for45 minutes before use.
- 3- Apply the sample spots, and the standard spot on the silica gel plates, on the base line.
- 4- Put the silica gel plate in the glass tank and allow the mobile phase to rise to about two-third the plate.
- 5- Remove the plate from the tank, and allow drying, and then detecting the spots by the use of the spray reagent.
- **Result:** Orange spot will appear.

