(Home works lecture 2)

Q2: Calculate the activity of 137 Cs after: (a) 2 years, (b) 15 years, (c) 30 years; if its activity at production was (10 μ Ci) and the half – life of 137 Cs is (30 years)?

 $A_o = 10 \,\mu\text{Ci}$ (Activity at production)

$$t_{1/2} = 30 \text{ y}$$

(Activity at t) $A = A_0 e^{-\lambda t}$

$$\lambda = \frac{0.693}{t_{1/2}} = \frac{0.693}{30 \text{ y}} = 0.0231 \text{ y}^{-1}$$

(a) (At t = 2 y) A= 10
$$\mu$$
Ci × e^{-0.0231 y-1 × 2 y} = 9.5 μ Ci

(b) (At t = 15 y) A= 10
$$\mu$$
Ci × e^{-0.0231 y-1 × 15 y} = 7 μ Ci

(c) (At t = 30 y)
$$A = 10 \mu \text{Ci} \times e^{-0.0231 \text{ y} - 1 \times 30 \text{ y}} = 5 \mu \text{Ci}$$

Q3: Radioactive element was produced at 8 - 1 - 2020 with radioactivity 5 mci half-life of this source is 5.6 y. calculate its activity on today's date?

 $A_o = 5$ mCi (Activity at production)

$$t_{1/2} = 5.6 \text{ y}$$

$$\lambda = \frac{0.693}{t_{1/2}} = \frac{0.693}{5.6 \text{ y} \times 365.25} = 0.00034 \text{ day}^{-1}$$

$$t = (26 - 2 - 2024) - (5 - 1 - 2020) = 21 \text{ day} + 1 \text{ month} + 4 \text{ year}$$

$$t = 21 \text{ day} + 1 \times 30 \text{ day} + 4 \times 365.25 = 1512 \text{ day}$$

$$A = A_0 e^{-\lambda t}$$

A= 5 mCi × e $^{-0.00034 \text{ day-1}} \times ^{1512 \text{ day}} = 5$ mCi × 0.59 = 2.9 mCi

(Home Work lecture 4)

Examples 1: Gamma photon emitted from the cesium source 137 Cs, has energy 0.662 μ eV, and was absorbed with the electron in the ground level of the hydrogen atom by the photoelectric interaction. Hydrogen binding energy 13.6 eV, calculate the kinetic energy of the electron?

فوتون كاما المنبعث من مصدر السيزيزم 137 Cs وطاقته $^{0.662}$ µev امتصت مع الالكترون في المستوي الارضي لذرة الهيدروجين تفاعلا كهروضونيا طاقة ربط الهيدروجين $^{13.6}$ eV احسب الطاقة الحركية للالكترون المتحرر؟

Answer:

$$hv = E_R + E_K$$

Where: hv = Energy of incident photon.

 E_B = Binding energy of electron.

 E_K = Kinetic energy of electron.

$$E_K = hv - E_B$$
 $E_k = 0.662 \times 10^6 \text{ eV} - 13.6 \text{ eV} = 661986.4 \text{ eV} = 0.6619 \,\mu\text{ eV}$

Example 2: Calculate the energy, frequency and wavelength of the scattered photons at an angle $\Phi = 90^{\circ}$ when the energy of the incident photons is 1.173 μeV ; Calculate the kinetic energy of the outgoing electron?

احسب الطاقة والتردد والطول الموجي للفوتونات المستطارة بزاوية $\Phi=90$ عندما تكون طاقة الفوتونات الساقطة $1.173~\mu~eV$ ثم احسب الطاقة الحركية للالكترون المتحرر؟

Answer:

$$h\upsilon^{-} = \frac{h\upsilon}{1 + \frac{h\upsilon}{m_0c^2}(1 - \cos\Phi)}$$

$$hv^{-} = \frac{1.173}{1 + \frac{1.173}{0.511}(1 - \cos 90)}$$

$$h\nu^- = 0.355 \ \mu \ eV = 0.355 \times 10^6 \times 1.6 \times 10^{-19} = 0.568 \times 10^{-13} \ J$$

$$\mathbf{E}^{-} = h v^{-}$$
 $0.568 \times 10^{-13} \, \mathbf{J} = 6.6 \times 10^{-34} \, \mathbf{J.S} \times v^{-}$ $\sim v^{-} = 0.0856 \times 10^{21} \, \mathrm{sec^{-1}}$

$$C = \lambda^{-} \times v^{-} \lambda^{-} = C / v^{-} * \lambda^{-} = 3 \times 10^{8} \,\text{ms}^{-1} / 0.0856 \times 10^{21} \,\text{sec}^{-1} = 35.04 \times 10^{-13} \,\text{m}$$

$$hv = hv^- + E_K$$
 $E_K = hv - hv^- * E_K = 1.173 \ \mu \text{ eV} - 0.355 \ \mu \text{ eV} = 0.818 \ \mu \text{ eV}$

Example 3: Calculate the kinetic energy of the positron resulting from the pair production interaction when the energy of the incident photon is 2.022 µeV?

احسب الطاقة الحركية لبوزترون الناتج من تفاعل انتاج الزوج عندما تكون طاقة الفوتون الساقط $2.022~\mu~eV$?

$$hv = m_{o_e}C^2 + m_{o_e}C^2 + E_{K_{e-}} + E_{K_e}$$

$$hv = 2 m_0 C^2 + 2 E_k$$
 $2.022 = 2 \times 0.511 + 2 E_k$

$$2 E_k = 1$$
 $\star E_k = 0.5 \,\mu \,\text{eV}$

Example 4: Calculate the thickness of the water layer that reduces the number of photons to 80% of its original number? Where $(\mu_m)_{\text{water}} = 0.0706 \text{ cm}^2/\text{g}; \rho = 1 \text{g/cm}^3$

$$(\mu_m)_{\rm water}$$
 وحسب سمك طبقة الماء التي تقلل عدد الفوتونات الى %80 من عددها الاصلي حيث $ho=1 {
m g/cm}^3$

$$I = I_o e^{-\mu x}$$

$$\mu_m = \mu_l / \rho$$
 $\mu_l = \mu_m \times \rho$ $\mu_l = 0.0706 \text{ cm}^2/\text{g} \times 1 \text{g/cm}^3 = 0.0706 \text{ cm}^{-1}$

$$I = 0.8 I_{\rm o}$$

$$0.8 I_o = I_o e^{-0.0706 x}$$

$$\text{Ln } 0.8 = -0.0706 x$$

$$x = 3.16 cm$$