

*Lecture3***Shell model****Advantages of the shell model:**

1. The **shell model** was able to explain the **existence of magic numbers** (2, 4, 20, 28, 50, 82, 126).
2. The **shell model** was able to provide correct values for the **angular momentum** of the nuclei in their ground state.

Disadvantages of the shell model:

1. The **shell model** failed to predict the **excitation states** of the nucleus.
2. The **shell model** failed to explain the **spherical asymmetry** of many nuclei.

Theory of The Nuclear Shell Model

The **nucleons** inside the nucleus are **linked** in the form of **orbits** similar to the orbits of electrons of the atom, and this has been called the **shell structure** or the **structure of levels** in which some of the shells are **closed** due to the **stability** of some nucleons.

The scientists **W. Elasser** and **M. Mayer** proposed the first idea of the **closed nuclear shell**, where the **total angular momentum of the nucleus** was understood through this model, so that the **total angular momentum** of any nucleus was formed from the **spin of its components**:

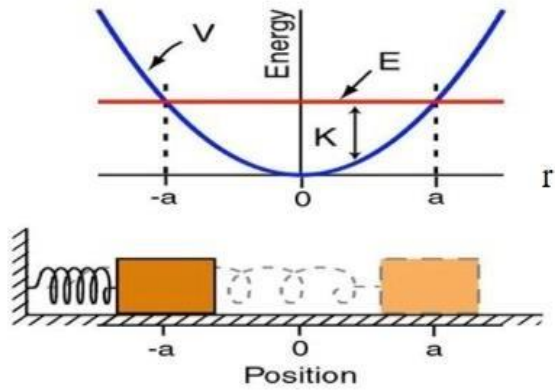
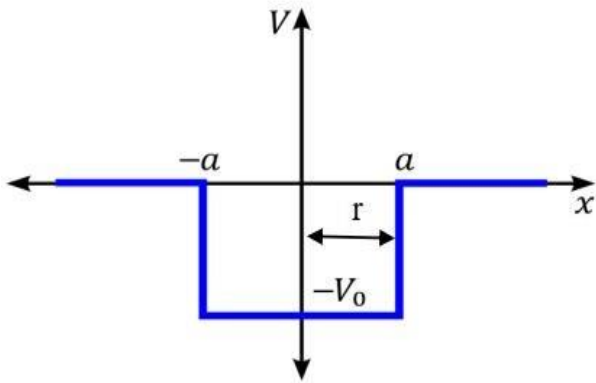
1. The nuclear spin of **protons** equal to $(\frac{1}{2}\hbar)$.
2. The nuclear spin of **neutrons** equal to $(\frac{1}{2}\hbar)$.
3. **Orbital Angular Momentum** of nucleons is due to its motion in the nucleus.

The **structure of the nuclear shell** is not easily reached, due to our lack of knowledge of the **nuclear potential**. So, through the **characteristics** of the **nuclear force**, the **nuclear potential** can be imposed, which **depends on** two basic **assumptions**:

1. Each **nucleon** moves **freely** and **fluently** in the **field** which is called **potential** which is the diagonal distance from the center of the nucleus.
2. The **energy levels**, or **shells**, are **filled** relation to the Pauli Exclusion Principle.

Based on these two assumptions, many solutions and calculations have been made according to quantum mechanics to develop a **general model** that includes the presence of the shell or nuclear levels, using the **potential models** as:

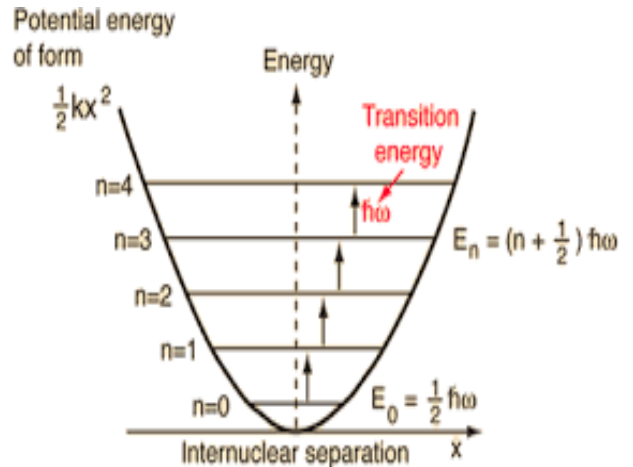
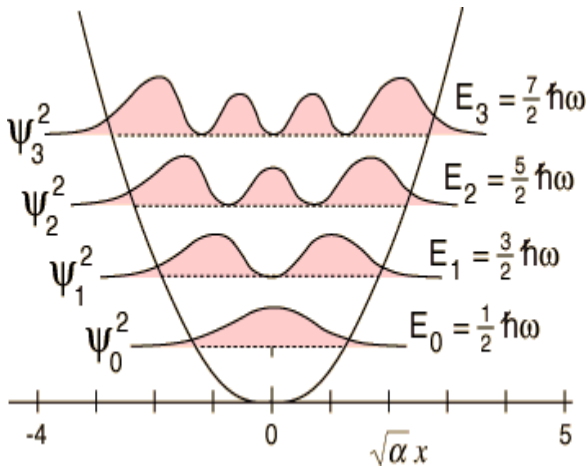
1. **Square Well Potential** (the nucleus has **spherical** shape).
2. **Harmonic Oscillator Potential**.



Square Well Potential

$$\left. \begin{array}{l} x \leq r \quad V(x) = -V_0 \\ x > r \quad V(x) = 0 \end{array} \right\} \text{حاجز الجهد}$$

$$\left. \begin{array}{l} x < r \quad V(x) = 0 \\ x > r \quad V(x) = \infty \end{array} \right\} \text{حاجز غير محدود}$$



For **Harmonic Oscillator Potential**:

$$V(r) = -V_0 + \frac{1}{2}Kr^2 \quad , K = m_0\omega^2, \quad \omega = \sqrt{\frac{k}{m}}$$

Where ω is frequency and m_0 is mass of particle and K is wave number

There is a **model** linking between **Orbital Angular Momentum (L)** and **Spin Angular Momentum (S)** for each nucleon and it is called **Spin- Orbital Coupling Model**. In this **model** the nucleons move inside the nucleus in the potential field of the nucleus and there is no possibility of collision with each other according to **Pauli's rule**.

The **Nuclear orbits** of the nucleons are split according to the following equation:

$$J = l \pm \frac{1}{2}$$

Where: l is the **number of orbitals** (0, 1, 2, 3, 4, 5...), and J is the **angular momentum**.

The **Nuclear orbits** are **filled in** according to the following equation:

$$\text{number of nucleons} = 2J + 1$$

- The **total angular momentum** of any nucleus in the ground state that contains its nucleons (**even - even**) is equal to **zero**, meaning that:

$$\sum J_n = 0 \quad , \quad \sum J_p = 0$$

Where: J_n and J_p are the total angular momentum of neutrons and protons respectively.

- In a nucleus that contains numbers of **even neutrons** and **odd protons**, the **total angular momentum** with respect to the neutrons is equal to

