

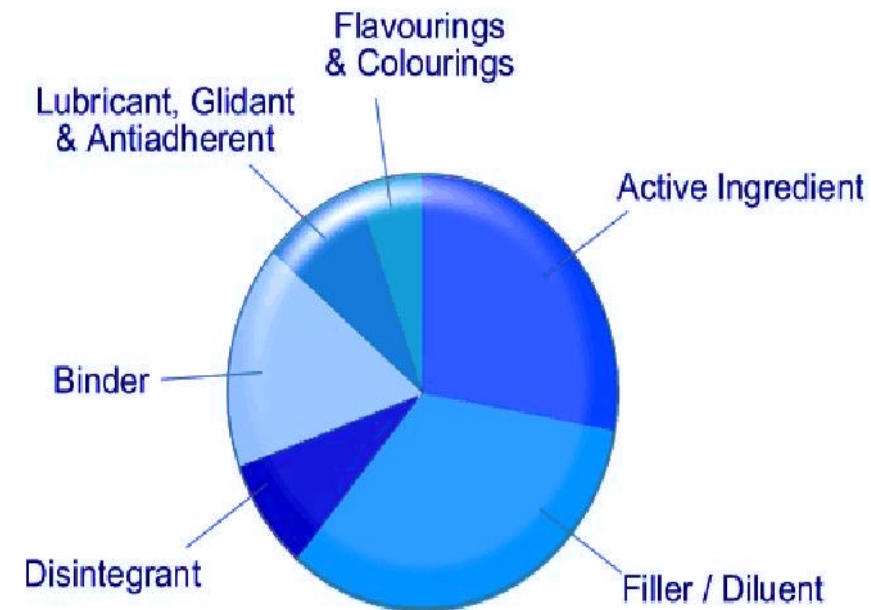




## Lec. 2: Excipients

# Tablet Excipients

- Tablets usually contain materials in addition to the active ingredient.
- All "nondrug" materials of the formula are called **excipients**.
- Excipients are **necessary** for the following reasons:
  1. Some drugs have **small doses** that cannot be compressed alone, e.g., Digoxin 0.25 mg.
  2. Some drugs have **poor compressibility** and flowability that cannot be compressed alone (e.g. Metformin).
  3. If the tablets are **compressed alone**, → they will not disintegrate **or** disintegrate very slowly (→ excipients help in disintegration and dissolution).



**Typical Tablet content**

# Tablet Ingredients (From Lec. 1)

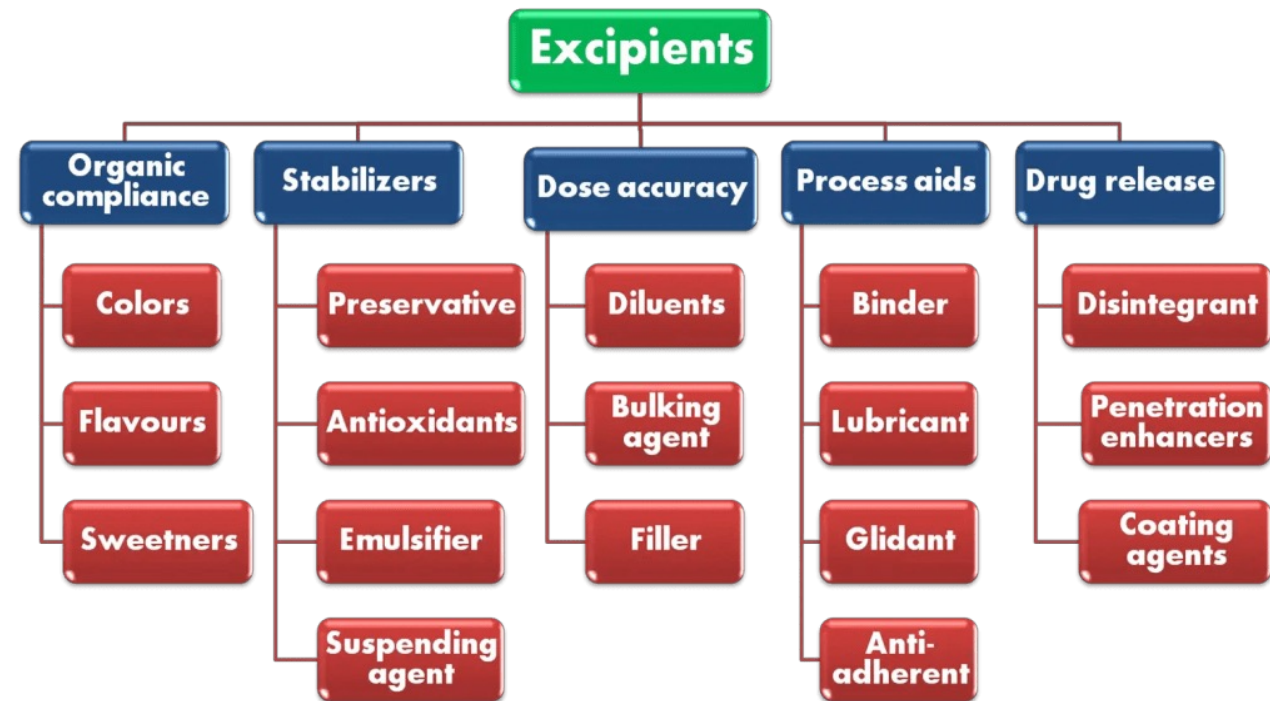


- **Properties of tablet ingredients:**
  1. **Nontoxic** and legal in the countries where the product is to be marketed.
  2. Must be commercially **available**.
  3. Reasonable **cost**.
  4. Must not be **contraindicated** by themselves (e.g. sucrose) or because of the component (e.g. sodium) in any segment of the population.
  5. Must be physiologically **inert** (for excipients)
  6. Must be physically and chemically **stable** themselves or in combination with the drugs(s) and other tablet component.
  7. Free of unacceptable **microbiological** contaminations.
  8. Must be **color compatible** (must not produce any off-color appearance).
  9. Must **not alter bioavailability** (for excipients)

# Types of Excipients

• The main excipients used in tablet formulation are:

1. **Diluents (fillers)** such as lactose, starch, and microcrystalline cellulose.
2. **Binders (granulating agents)** such as starch, acacia, and gelatin.
3. **Disintegrants** such as starch, and super disintegrant.
4. **Lubricant, Antiadherent, Glidants** such as magnesium stearate and stearic acid.

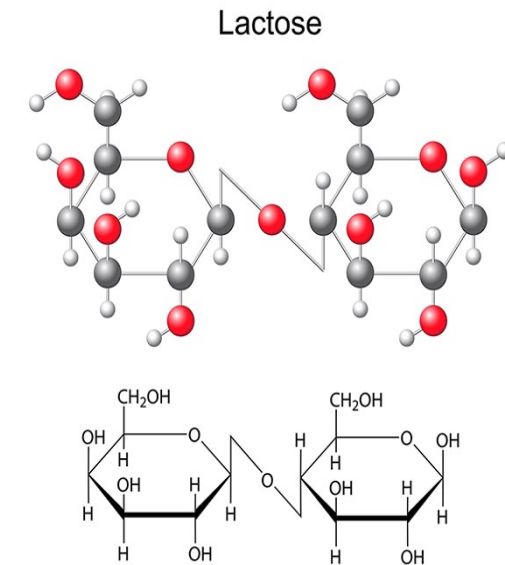


# Diluents (Fillers)

- Diluents are materials used to **make up the required bulk** of the tablet when the drug itself is inadequate to provide this bulk.
- **Occasionally**, the active ingredient has a **large dose and good compressibility** so that it does not need diluent, e.g., aspirin and some antibiotics.
- **However, most tablets need a diluent.**
- Round tablets are usually in the size range of **5-13 mm**. Tablets **below** 5mm may be difficult for the elderly to handle and those **larger** than 13 mm become difficult to swallow.
  1. Diluents are therefore used to formulate the tablets within the desired size range.
  2. Diluent also can **have a secondary** advantage which is to provide better tablet properties such as improved **cohesion** or to **promote flow**.
- There are large numbers of diluents and the following are examples of them:
- Lactose, Starch, Dextrose, Mannitol, Sucrose, Microcrystalline Cellulose

# Lactose

- The **most widely used** diluent in tablet formulation has no reaction with most drugs.
- There are **three** forms of lactose available( anhydrous, hydrous, and spray-dried).
- **Anhydrous** lactose has **an advantage** over the other two types; on aging, it does **not** undergo discoloration (brown discoloration, Maillard reaction) with **amines and alkali** compounds.
- The **anhydrous** lactose may pick up **moisture** from the environment when exposed to elevated humidity and convert to the hydrous form.
  - → These Tablets should be **carefully packaged** to prevent moisture exposure.
- When **wet granulation** is used for the production of tablets, the **hydrous** form is usually used.



# Lactose

- In general, all lactose types show the following **advantages**:
  1. Good drug **release**.
  2. In granulations, granules are **easily dried**.
  3. The **disintegration time** of lactose-containing tablets is **not very sensitive** to variations in tablet hardness.
  4. Low **cost**.
  5. Has **no reaction** with most drugs.
- **Disadvantages**: **Hydrous** form undergoes discoloration when used with alkali or amine-containing compounds

Note: The other type of lactose is:

- **Spray-dried** lactose is used for **direct compression** due to its good **compressibility and flowability**.



# Starch

- It may be derived from different sources such as corn, wheat, or potatoes.
- Great **care** should be taken when using starch in the formula **because** it can be used as **diluent**, **binder**, or **disintegrant** depending on:
  1. **Type of starch:** the useful type for a particular formula can be known by experts.
  2. **The amount used:** it is used as a **diluent** in the ratio of 50-60%, **binder** in the ratio of 2-10%, and **disintegrant** in the ratio of 5-20%.
  3. **Stage of addition:** it is used as a **diluent** when added in the **dry form** at the beginning of the procedure (mixing step), a **binder** when used **as a paste** in the preparation of the wet mass step, and **disintegrant** when added finally after granulation **as a dry form**.

# Other Diluents

**Dextrose:** It is available in two forms: **hydrous** and **anhydrous**.

- Dextrose is sometimes used in the formulas to replace some of the lactose to minimize the discoloration (when used with alkaline compounds).

**Mannitol:**

- It is widely used in **chewable and orodispersible** tablets because of its **sweet taste** (sugar) **pleasant feeling** in the mouth (due to the negative heat of solution), and **slow solubility**.
- It is **non-hygroscopic** so can be used safely in water-sensitive formulations like vitamin formulations.
- However, it is somewhat **expensive**, has **poor flow**, and **requires a high lubricant level**.

**Sucrose:** It is used sometimes in tablet formulations.

- In general, sucrose and sucrose-based diluents are avoided by some manufacturers due to their effect on diabetic patients.

## Microcrystalline cellulose:

- It is often referred to by the trade name **Avicel**<sup>®</sup>.
- Different grades of Avicel are available such as PH 101 (powder) and PH 102 (granules).
- It is a multipurpose excipient used as a **diluent** and **disintegrant**.

## Advantages:

- It is **inert** and can be used with alkaline or acidic substances (No discoloration).
- Has high purity and low moisture content.
- Avicel is a **directly compressible** diluent due to its **good compressibility** and **flowability**.
- Avicel-containing tablets are characterized by short disintegration time, high hardness, low friability, and low weight variation. (*Why?*)

# Binders (Granulating agent)



- They are substances that **bind the particles together** to form granules (in wet and dry granulation) or to promote the formation of cohesive compacts (in Direct compression). Below are some examples:

## Acacia and tragacanth:

- **Natural** gums. These materials are more effective when added as a solution than if they are used as powders.

## Disadvantages:

- They are variable in their composition (**why?**) and performance according to their origin.
- They are also **easily contaminated by bacteria**.

## Gelatin:

- **Synthetic protein** is preferred over acacia and tragacanth and is also easier to prepare in solution than the two gums. However, bacterial growth is also troublesome.

## Starch:

- One of the **most commonly used** granulating agents (binder) and used as a **paste**.
- It is prepared by dispersing starch into water which is then **heated** for a certain time in order to induce starch hydrolysis into dextrin and glucose.
- A properly made paste is **translucent rather than clear** (which indicates complete conversion to glucose).

## Modified Natural Polymers:

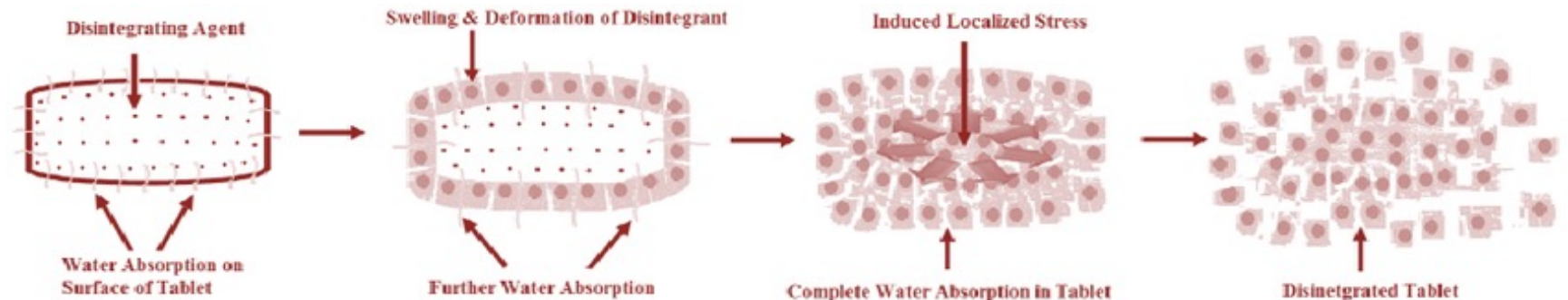
- Common and important binders. Alginates (e.g., **sodium alginates**) and cellulose derivatives (e.g., **methylcellulose (MC)**, **ethyl cellulose (EC)**, **hydroxypropyl cellulose (HPC)** and **hydroxypropyl methylcellulose (HPMC)**) are examples of these binders.

# Binders

- **Modified Natural Polymers (continue):**
- **Except for EC**, all of the cellulose derivatives can be used as dry powders (in Direct compression and dry granulation) and as an aqueous solution (in wet granulation).
- **HPC** can also be used as an **alcoholic solution**, thus it is useful for water-sensitive drugs.
- **EC** is used **only** as an alcoholic solution because it is **insoluble in water**, therefore it may **retard** tablet disintegration.

# Disintegrants

- A disintegrant is a substance that **facilitates the breakdown** of the tablet into smaller fragments upon contact with GI fluids. (<https://youtu.be/s5aGmUQIzSs> )
- The function of the disintegrant is to **oppose the effect of the tablet binder** and the **physical force** that is applied during the compression process.
- The disintegrants **act by** drawing water into the tablet, swelling and rupturing the tablet.
  - This tablet fragmentation is critical to the **subsequent dissolution** of the drug and to achieve **satisfactory bioavailability**.
- Disintegrants may added **at two stages**: during the formation of granules (to give intragranular action) and at the second mixing stage during compaction of granules into tablets (extragranular).



# Disintegrants



## Starch

- The **most commonly used disintegrants** because of their **low cost**. It is used in the ratio of 5-20% of the tablet weight.
- Starch has the property of **rapid water uptake and swelling** that leads to the rupture of the tablet due to the increase in internal pressure.

## Super Disintegrant:

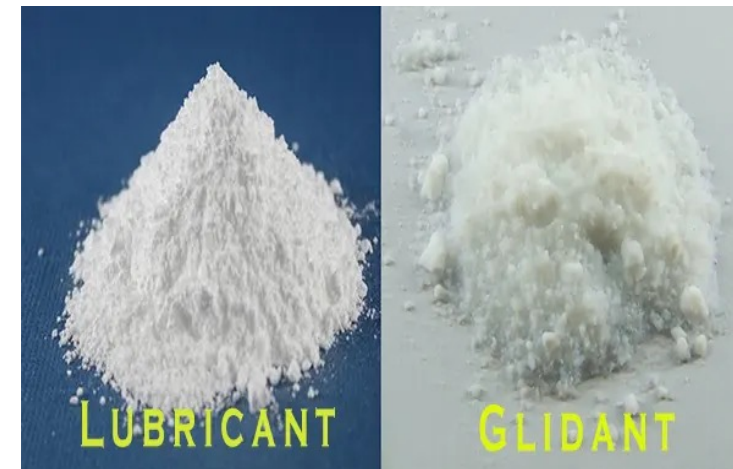
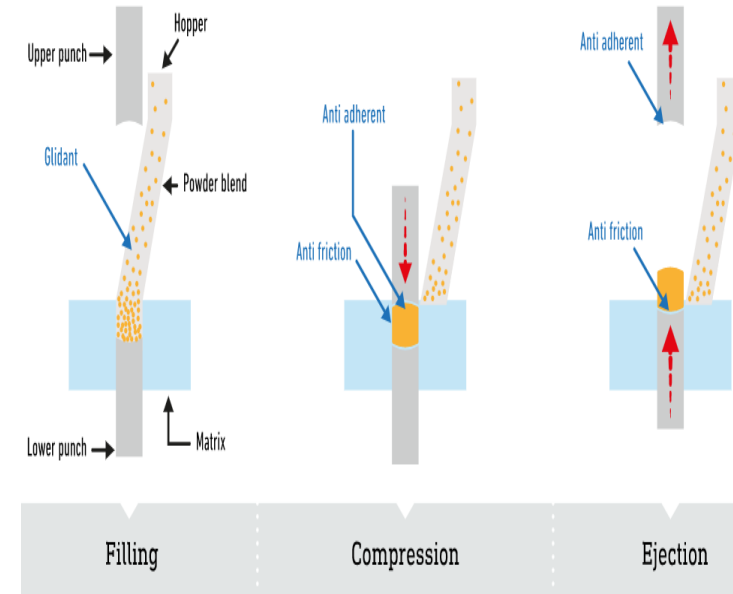
- They are so-called due to their powerful disintegrating action. Examples of these materials are **sodium starch glycolate (Explotab®)**, **croscarmellose sodium**, and **crospovidone (Kollidon CL)**.
- They are **very potent** if compared with the classic disintegrants. For example, croscarmellose sodium swells to **900% of its original** volume in acidic media while starch swells to 25% only in the same media. They are used when rapid disintegration is required such as in orodispersible tablets.



# Lubricants, Antiadherent, and Glidants



- **Lubricants:**
  - They are materials used to **reduce the friction** during tablet ejection **between the tablet and the walls of the die cavity** in which the tablet was formed.
- **Antiadherents:**
  - Reduce sticking of tablet granules to the **faces of the bunches** or to the die wall
- **Glidants:**
  - They are used to **promote the flow** of granules or powders by reducing the friction between the particles themselves.
  - Glidants are thought to work by **filling irregularities** in granules making them more round and reducing friction between granules.



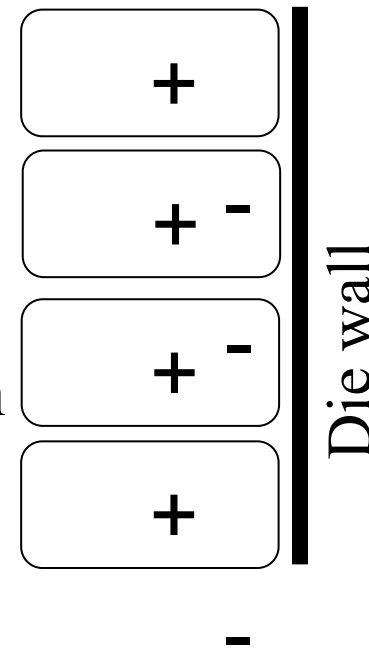
# Lubricants

- **Advantages** of lubricants

1. **Facilitate** tablets' ejection and prevent their sticking in the die.
2. **Prolong** the life of the die.
3. **Decrease** the liberated heat (friction heat).

- **Mechanism of action of the lubricants**

1. **Fluid lubrication** (Hydrodynamic (formation of thin film)): this mechanism is used to explain the action of **liquid** lubricants.
  - In general, all liquids (especially oily ones) decrease the friction between two surfaces.
- **Example: Mineral oils** such as **liquid paraffin** have been applied on the granules as a **fine spray**.
  - **However**, the **problem** with using this type of lubricant is the **production of oil spots**.
  - Another problem is that the **particle surface** will be hydrophobic and the tablet may have a **slower dissolution** rate → may alter bioavailability.



- **Mechanisms of lubricants (continue):**

2. **Boundary lubrication**: this mechanism is used for **solid** lubricants.

- In this type, the **polar portion** (such as  $-\text{OH}$ ,  $-\text{NH}_2$ ) of the lubricant is attached to the metal and prevents the tablet from sticking to the die.

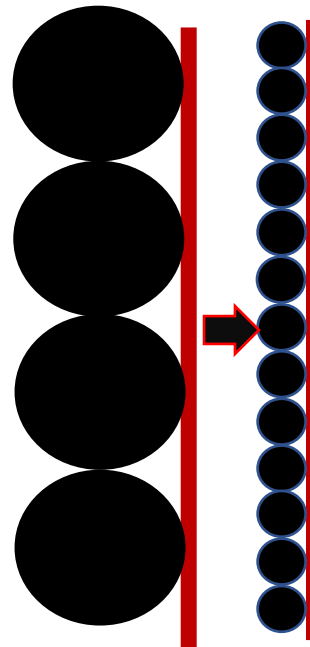
- The lubricants should be added in the **last step** (just before compression) since they must be **present on the surface** of the granules and not between them.



# Notes About Lubricants



1. The **particle size** of the lubricants is crucial; they should be 200 mesh in size or finer.
  - As a general rule, as the particle size of the lubricants **increases**, their efficacy **decreases**.
2. The **amount** of the lubricant in the formula should **not exceed 1%** (for most lubricants) and this is due to the following problems that can happen with increasing lubricant amount:
  - These materials are **water-insoluble** and present on the surface of the granules, thus retard water penetration and **decrease dissolution** rate and may alter bioavailability.
  - Adding excess lubricants will result in the hydrocarbon portion covering the whole surface of the tablet and the interaction between particles will be **replaced by** the interaction of hydrocarbons which is **weaker** and thus the tablet structure is **weakened**.



3. The **mixing time** of the lubricant with the formula should be 2-5 min.
  - **Over-mixing** decreases the lubricant efficacy because it causes the penetration of the lubricant from the surface to the core of the formula.
4. The **mixing rate** is also important; a high mixing rate causes the penetration of the lubricant inside the core of the formula and thus, decreases the lubricant's efficacy.
- **Examples** of the commonly used lubricants:
  - **Magnesium or calcium stearate** are the **most** widely used lubricants due to their efficacy. These lubricants should **not** be used with acidic drugs like aspirin (due to pH-induced hydrolysis ).
  - **Stearic acid** is less effective than its magnesium and calcium salts. It should **not** be used with alkaline drugs.

# Lubricants and Glidants



- **Zinc stearate** is inert with good lubricating properties and **small particle size**. It is used effectively in **direct compression**.
- **Talc** may also be used as a lubricant. **Problem:** it does contain a trace amount of **iron**, so it should be applied carefully in any formulation containing a drug whose breakdown is catalyzed by the presence of iron.
- **Regarding the glidants**, they are less commonly used than the lubricants. Examples of these materials are **talc** and **Aerosil®**.

# Colorant



- These materials are used for the following reasons:

1. Product **identification**.
2. The production of more **elegant products**.
3. To **hide undesirable properties**. For example, vitamin C undergoes oxidation that causes brown discoloration, **therefore** vitamin C tablets are usually colored with yellow or red color.



- **Types of colorants**

1. **Natural colorants**: are **not** preferred because they are limited (few colors are available) and **unstable**.
2. **Dyes**: are synthetic colorants applied as **solution**. They are dissolved in the granulating solution. When using water-soluble dyes, care should be taken to prevent their migration during the drying process.
3. **Lakes**: they are **powders**, **insoluble** in water mixed with the formula (as powder), and compressed with the tablet to give the desired color.

# Flavors

- They are usually **limited to chewable** tablets or other tablets intended to dissolve in the mouth to mask the undesirable taste.
- It is usually added in the percent of 0.5-0.75%.
- There are different types of flavors:
  1. **Water soluble flavors:** they are **not** preferred due to their poor stability.
  2. **Oily flavors:** they dissolved in a suitable solvent and **mixed with the binder** solution. It is important that such flavors should tolerate the subsequent drying.
  3. **Dry flavors:** these are dry powders blended with the formula.



# Sweeteners



- They are also used in mouth tablets.
- **Sucrose** is **not** used commonly due to its effect on patients with diabetes.
- **Mannitol** which is about 72% as sweet as sucrose can be used.
- Synthetic sugars such as **saccharine** are also used. This material is about 500 times sweeter than sucrose. However, it has the disadvantage of **bitter aftertaste** and it is reported to be carcinogenic on large doses.
- **Aspartame** is another synthetic sugar and is **more preferred** than saccharine. Its **disadvantage** is that it has **poor stability** in the presence of moisture.

# Tablet Design and Formulation

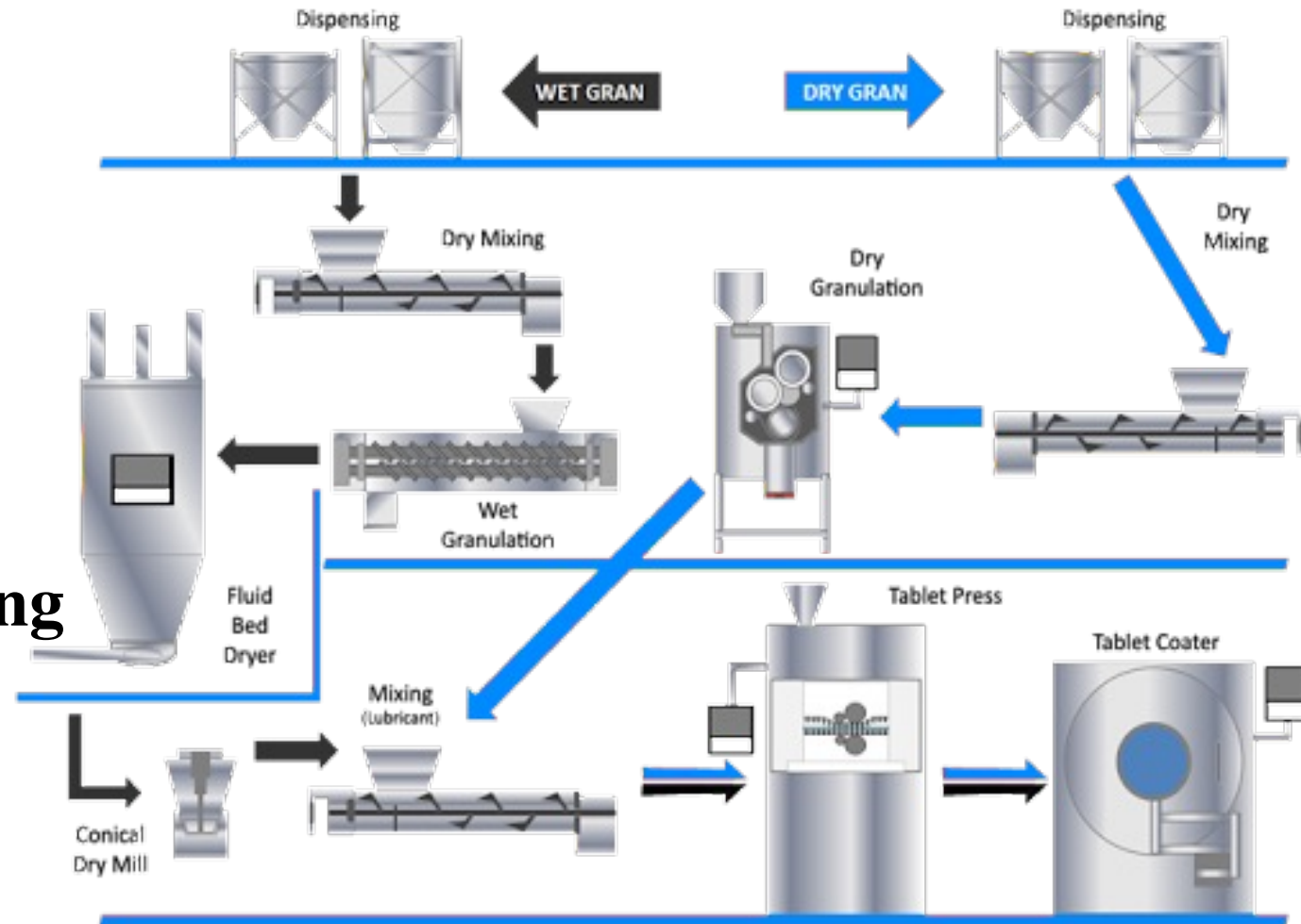


- Generally, tablet formulation consists of the following main processes:

1. Weighing.
2. Granulation.
3. Milling and mixing.
4. Compression.

- **Three types of tablet manufacturing process:**

1. Wet Granulation.
2. Dry Granulation.
3. Direct compression.



# Mixing

- A crucial step in the formulation of the pharmaceutical dosage form.
  1. Mixing allows **uniform distribution** of tablet constituents throughout the formulation
  2. Allows for **adequate distribution** of **lubricants** around tablet granules.
- **Undermixing** will probably lead to impaired **content uniformity** of tablet formulation and **poor flow** properties
- **Over-mixing** will lead to other tablet manufacturing problems such as **poor flow** properties due to lubricants entering into the tablet granules instead of staying on the surface.



**V- Shape Mixer**

# Granulation

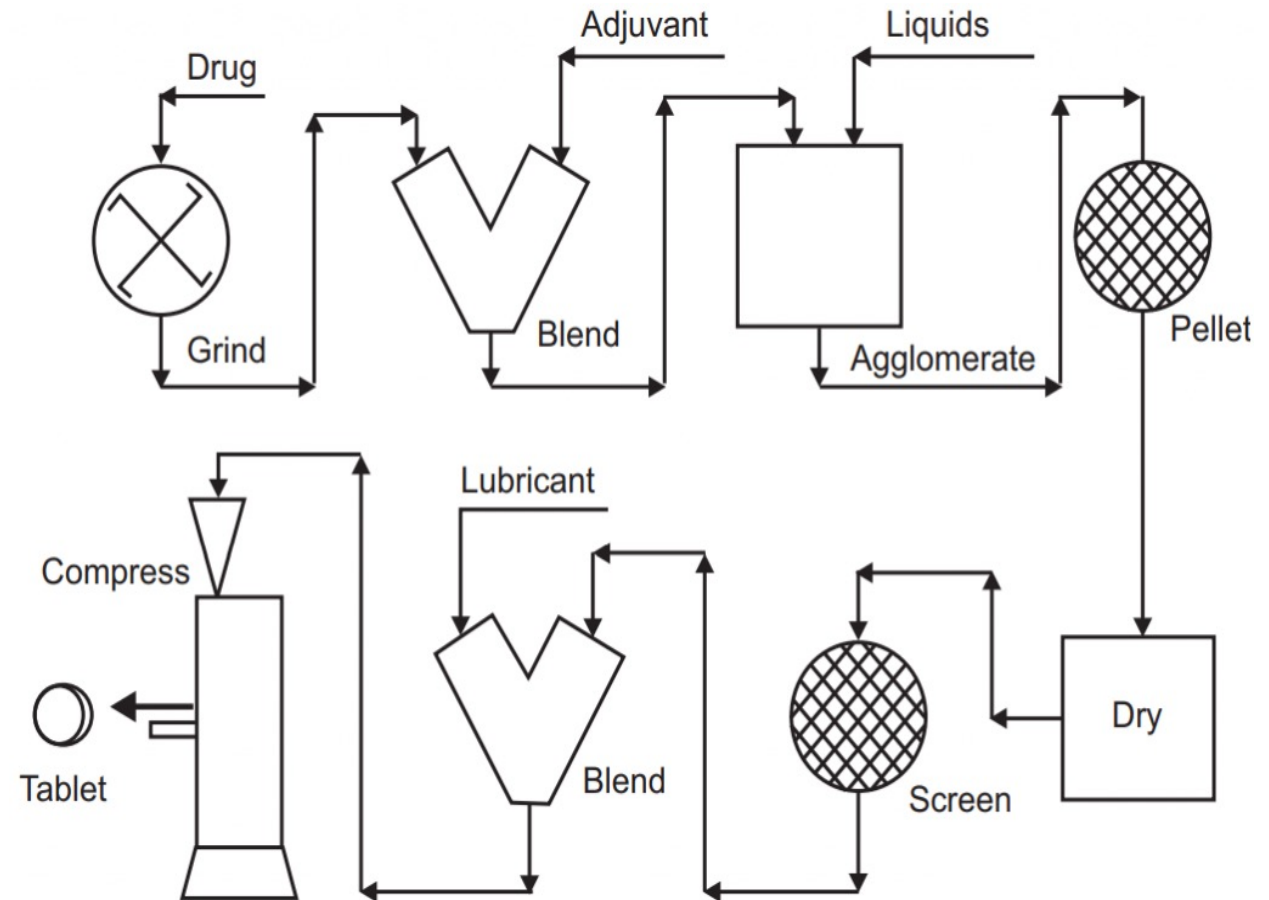
- Fine powder drug mostly has poor flow properties.
- Granules have:
  1. Better **flowability** than individual ingredients
  2. Better **compressibility** than individual ingredients.
  3. It ensures the **consistent spread** of API in the formulation.



# Wet Granulation (used liquid binder)



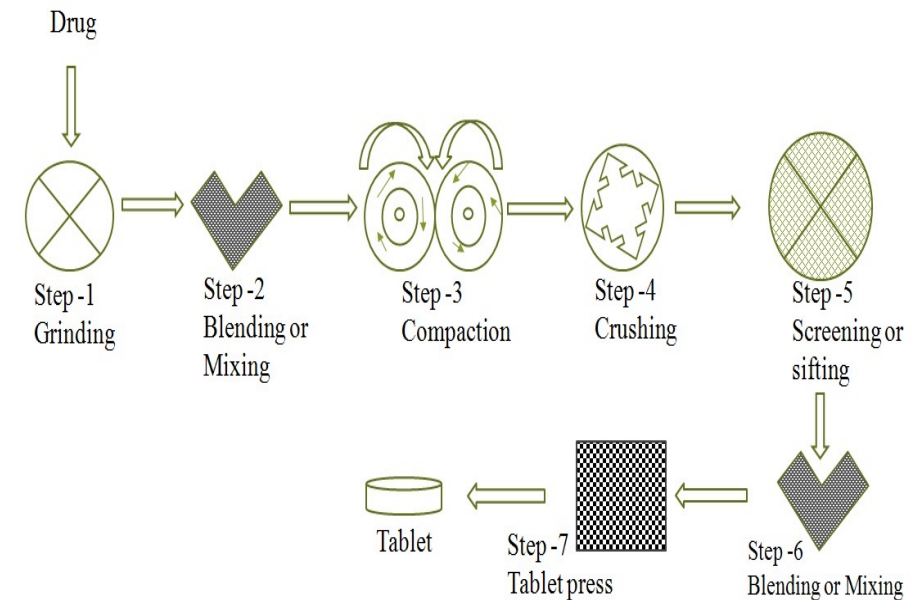
- **Most common**; Contains more steps but is relatively **easier** to control and compress.
- Active ingredients **plus** excipients are mixed together **then** a binder (binding **liquid**) is added to a rapid mixer granulator (machine)
- Granules are **then** dried **and** milled to the required size range
- **Then** mix with lubricant **and** compress



# Dry Granulation

- Used **when** there is a limitation for using a granulating solution such as the **interaction** between the active ingredient and the solution
- Also in cases when the drug is **sensitive to heat**.
  - Drug powder and inactive ingredients are mixed together and passed through a powder **roller compactor** machine to produce **slugs**.
  - This process is called slugging**
    - Slugs undergo milling and screening to get the desired particle size
  - Examples are vitamin formulations

Dry Granulation procedure



# Direct Compression



- **A special** case of tablet formulation.
- **No granulation step**; only mixing ingredients and excipients and compressing directly.
- It requires certain properties of the active ingredients to be able to produce a good tablet:
  1. **Crystalline** in nature.
  2. **Components are easy to compress.**
- Sodium chloride, potassium salts of iodide, or chloride tablets can be done using this method.
- **Limitation:** Most materials have weak intermolecular interaction attraction that tends to hinder compaction.

# Direct Compression



- **Advantages of Direct Compression:**
  1. Simple, low labor input and hence **economic**.
  2. Being a **dry process**, the risk of deterioration of the active ingredient is decreased.
  3. Tablets will **disintegrate onto their primary** particles rather than granular aggregates → the resultant increase in surface area available for dissolution should result in faster drug release.
- Note: in recent years and due to the development in machine technology and discovery of new excipients; this method is being used increasingly, especially for a drug that is effective in a **very low dose** because this method ensures “**better**” content uniformity than others

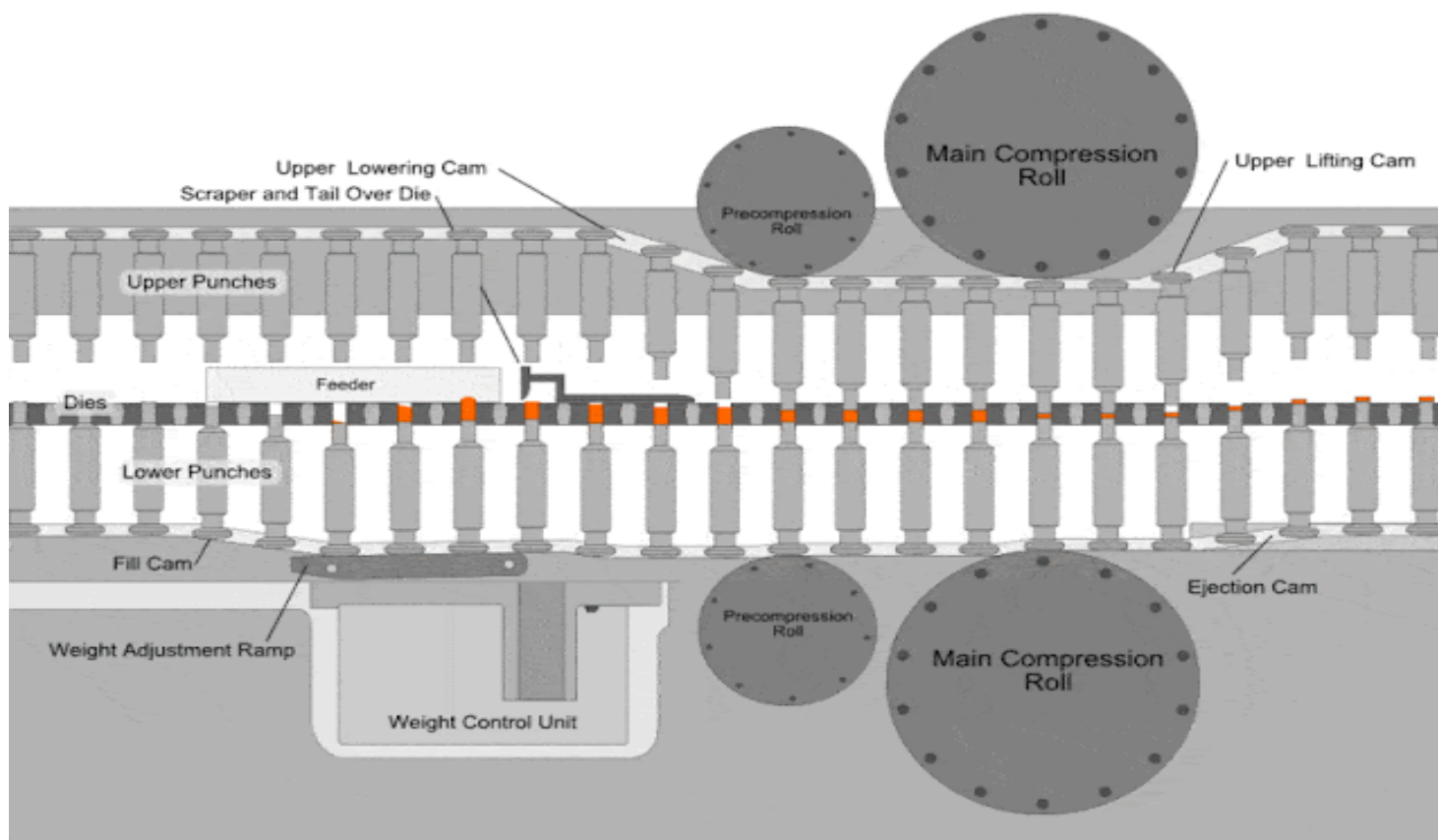


- **Disadvantage or limitation of D.C.:**

1. **Differences in particle size** and **bulk density** between the drug and diluent may lead to **stratification** within the granulation → may result in poor content uniformity of the drug in the compressed tablet.
2. A **large-dose drug** may present problems with direct compression if it is not easily compressible by itself. Because most drugs have low intermolecular forces and need other materials to increase interaction between molecules.
3. In some instances, the direct **compression diluent may interact** with the drug → may lead to problems such as **discoloration**.
4. Because of the dry nature of direct compression, **static charge buildup** can occur on the drug during routine screening and mixing which may **prevent a uniform distribution** of the drug in the granulation.

# Compression

- The final step is tablet formulation.
- Required a set of variables to be evaluated which are beyond the scope of this class.



# Tablet Compression Machine

1. **Hopper(s)** for holding and feeding tablet components.
  2. **Dies** that define the size and shape of the tablet.
  3. **Punches** for compressing the granulation within the dies.
  4. **Cam tracks** to guide the movement of the punches.
- **Note:** there are various auxiliary equipment designed to aid in tablet production such as automatic feeders, deduster (beyond the scope of this class)



