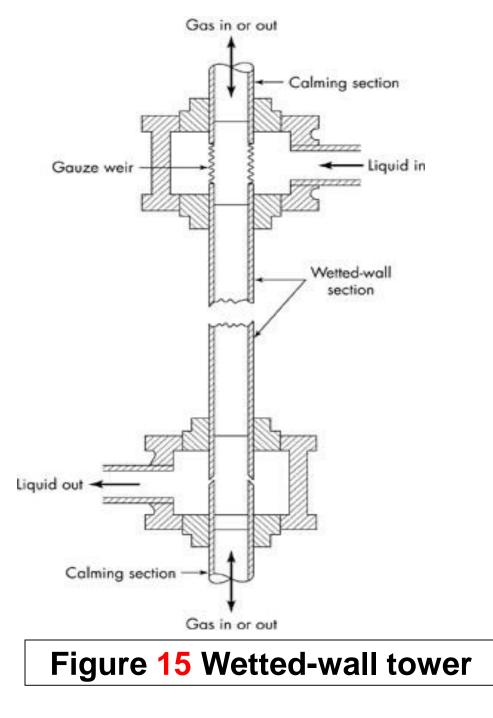
EXPERIMENTAL MEASUREMENT OF MASS TRANSFER COEFFICIENTS

- > The mass transfer coefficient, k_c , can be studied in experimental devices in which the area of the contact between phases is known.
- The <u>wetted-wall tower</u> is one of the instruments used in practice.
- It can give valuable information on mass transfer to and from fluids in turbulent flow.



- Generally, the gas enters the bottom of the tower and flows <u>countercurrent</u> to the liquid, but parallel flow (concurrent) can also be used.
- > Experiments were conducted for obtaining the numerical values of c_A or x_A . These values can then be used to calculate the mass transfer coefficient.
- Measurement of the rate of evaporation of the liquid into the gas stream over the known surface permits calculation of the mass transfer for the gas phase.
- The use of different gases and liquids provides variation of Schmidt number, Sc.

Mass transfer coefficient, k would depend on:

- 1) Diffusivity, *D*_{AB}
- 2) Velocity, v
- 3) Viscosity, μ
- 4) Density, ρ
- 5) Linear dimension, D

or:
$$k = \psi(D_{AB}, D, \upsilon, \mu, \rho)$$

Dimensional analysis gives:

$$\frac{k_c D}{D_{AB}} = \psi_1 \left(\frac{DG}{\mu}, \frac{\mu}{\rho D_{AB}} \right) = \text{Sherwood number, Sh} \qquad \dots (93)$$

where: $G = v\rho$

Mass transfer with flow inside pipes

Correlations for mass transfer to the inside wall of a pipe is given by Sherwood number, Sh:

a) For laminar flow (Re < 2100):

$$Sh = 1.76 \, Gz'^{1/3}$$
(94)

where:

Graetz number,
$$Gz' = \frac{\dot{m}}{D_{AB}L\rho} = \frac{\pi}{4} \operatorname{Re} \operatorname{Sc} \frac{D}{L}$$
(95)

* **Recall:**
$$\operatorname{Sc} = \frac{\mu}{\rho D_{AB}}$$
 $\operatorname{Re} = \frac{D \upsilon \rho}{\mu}$

b) For turbulent flow (Re > 2100):

Sh = 0.023Re^{0.8} Sc^{1/3}
$$\left(\frac{\mu}{\mu_w}\right)^{0.14}$$
(96)

The term ($\mu I \mu_w$) is usually about 1.0 for mass transfer, hence can be omitted from Eq. (96).

c) For flow inside wetted-wall towers:

Data for evaporation of several liquids in wetted-wall towers were correlated with slightly higher exponents for both Re and Sc numbers:

$$Sh = 0.023 Re^{0.81} Sc^{0.44}$$
(97)

A correlation for mass transfer at high Schmidt number (430 to 100000) was obtained by measuring the rate of solution of tubes of benzoic acid in water and viscous liquid:

$$Sh = 0.0096 Re^{0.913} Sc^{0.346}$$
(98)

Example :

- a) What is the effective thickness of the gas film for the evaporation of water into air in a 2-in.-diameter wettedwall column at a Reynold's number of 10,000 and a temperature of 40°C?
- b) Repeat the calculation for the evaporation of ethanol under the same conditions. At 1 atm, the diffusivities are 0.288 cm²/s for water in air and 0.145 cm²/s for ethanol in air.

Solution: For air at 40°C,

From Appendix 8 (McCabe, Smith and Harriott): $\mu = 0.0186 \text{ cP}$

$$\frac{\mu}{\rho} = \frac{1.86 \text{ x } 10^{-4} \text{ g/cm} \cdot \text{s}}{1.129 \text{ x } 10^{-3} \text{ g/cm}^{-3}} = 0.165 \text{ cm}^{2}/\text{s}$$

a) For the air-water system:

$$Sc = \frac{\mu}{\rho D_{AB}} = \frac{0.165}{0.288} = 0.573$$

From Eq. (97):

$$Sh = 0.023(10000)^{0.81}(0.573)^{0.44} = 31.3$$

In the film theory, from Eq. (77):

$$k_c = \frac{D_{AB}}{z_T}$$
 and from Eq. (93) $Sh = \frac{k_c D}{D_{AB}}$

Therefore:

$$Sh = \frac{D}{z_T}$$
 or $z_T = \frac{2.0 \text{ in.}}{31.3} = 0.064 \text{ in.}$

b) For the air-ethanol system:

Similar to a), determine the Sc and Sh numbers:

$$Sc = \frac{0.165}{0.145} = 1.14$$
$$Sh = 0.023(10000)^{0.81}(1.14)^{0.44} = 42.3$$

Finally, calculate the effective thickness of the gas film:

$$z_T = \frac{2.0 \, in.}{42.3} = 0.047 \, in.$$

Note that thickness, z_{τ} become smaller as the diffusivity decreases with mass transfer coefficient, k_c .