

ALMUSTAQBAL UNIVERSITY COLLEGE Iraq - Babylon



RENEWABLE ENERGY TECHNOLOGY
Sustainable Path For a Carbon Free Future

Refrigeration and Air conditioning Techniques

Engineering Department



Subject : Renewable Energy

Grade: 4th Class

Lecture :2 Introduction of Radiation

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RADIATION



Radiation :Mode of Energy transfer by electromagnetic waves
only mode to transfer energy without the presence of a substance
(fluid or solid). works best in a vacuum (empty space

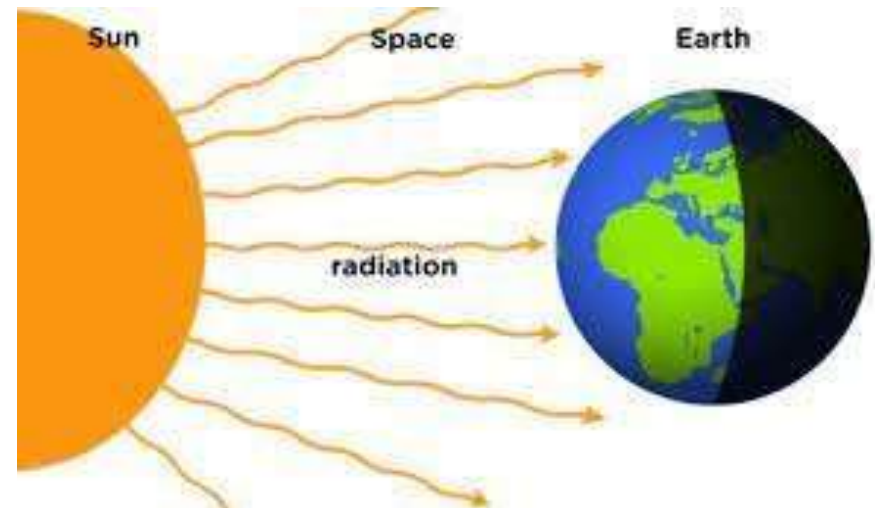
- Radiation = the **only way**
for Earth to receive
energy from the Sun

Weather systems are
powered by radiation

From Earth-Sun geometry we know:

- Spatial and temporal **variations** of receiving of radiation at
the top of the atmosphere

need to consider different types of radiation

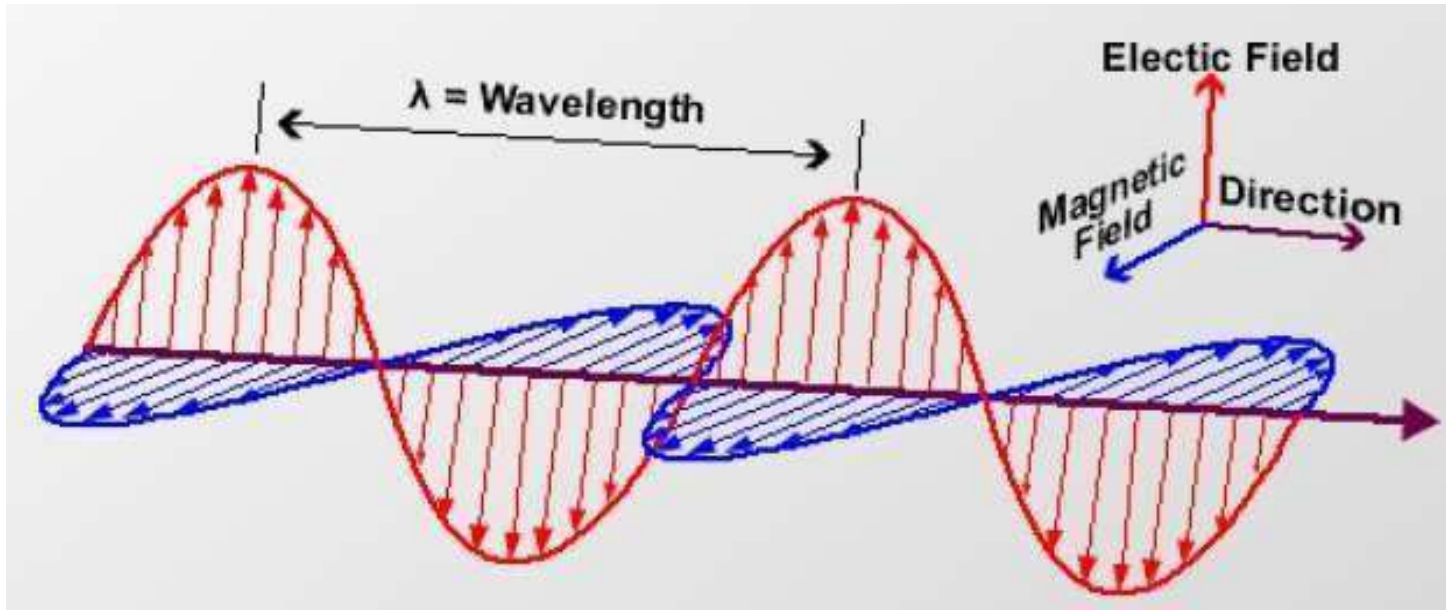


2.1 Electromagnetic spectrum



- **Radiation waves** exhibit characteristics of both **electric fields** and **magnetic fields**
- **Electromagnetic** radiation moves at “**speed of light**”
- Radiation spreads in **all directions** and moves in straight lines

Electromagnetic radiation is described by three **interdependent variables**:



wavelength λ "lambda" [m, μm]
 frequency ν "nu" [s^{-1} , Hz]
 velocity c [m s^{-1}]
 (c = "speed of light" $\sim 3 \times 10^8 \text{ m s}^{-1}$)

The relation between the λ , ν and c for each wave is :

$$\lambda \nu = c$$

2.2 Radiation Spectrum



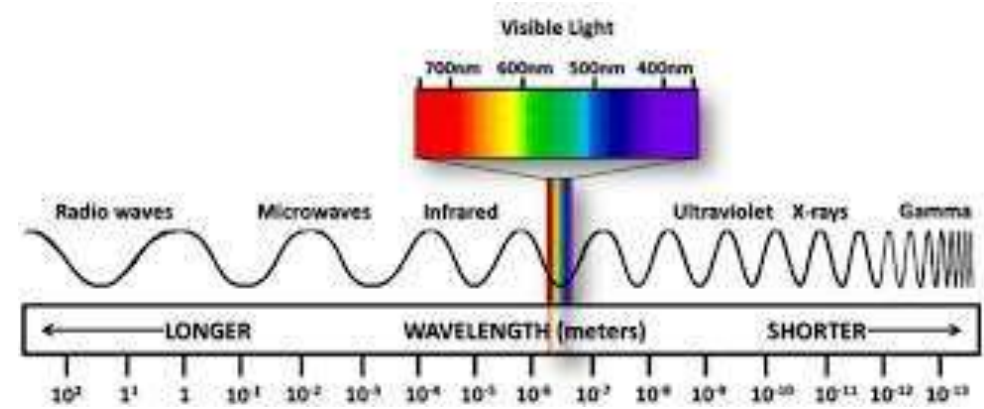
Definition:

The **Radiation Spectrum** is the **distribution** of **radiative energy** over different **wavelengths**, or frequencies.

In meteorology: only small part of EM- spectrum of interest.

➤ **three important ranges:**

- **ultraviolet radiation (UV)**
- **visible radiation**
- **infrared radiation (IR)**





2.2 Radiation Spectrum

Radiation in the Earth-Atmosphere System

	Ultraviolet Radiation UV	Visible Radiation	Infrared Radiation IR	
Wavelength	$10^{-2} - 0.4 \mu\text{m}$	$0.4 - 0.7 \mu\text{m}$	$0.7 - 100 \mu\text{m}$	
Effect	Sunburn	"sunlight"	heat-radiation	
		0.4 μm violet 0.5 μm blue green 0.6 μm yellow orange 0.7 μm red	near IR 0.7-1.5 [μm]	far IR 1.5 – 100 [μm]
Class				
sun output	7 %	43 %	37 %	11 %
Earth output	0 %	0 %	~0 %	~ 100 %

shortwave radiation: only solar radiation

longwave radiation: IR radiation emitted by the E/A-system

2.3- LAWS OF RADIATION



(iii) Reflection – Absorption – Transmission

part or all can be absorbed:

- fraction absorbed: **absorptivity**,
- this part **raises the temperature** of the object
- **radiative energy** is converted to **heat**

part or all can be transmitted:

- fraction transmitted: **transmissivity**, → this part does **not interact** with the object, it just goes through it.

Since these are the only possibilities, it follows from the principle of conservation:

$$\alpha_{\lambda} + a_{\lambda} + t_{\lambda} = 1$$

2.3- LAWS OF RADIATION



(iv) Stefan-Boltzmann Law:

the total emitted energy flux

All objects or substances emit radiation at a rate proportional to the 4th power of their absolute temperature

Total energy flux emitted: F_{tot} [W m⁻²]

:

$$F_{\text{tot}} = \varepsilon \sigma T^4$$

ε emissivity (0 ~ 1); depends on quality of material .

σ Stefan-Boltzmann constant = 5.67×10^{-8} [W m⁻² K⁻⁴]

T absolute temperature of emitting object [K]

T^4 fourth power: faster than linear increase with temperature.

2.3 LAWS OF RADIATION



Example:1

If a cloud bottom has a temperature of $-10\text{ }^{\circ}\text{C}$, how much energy would it be emitting if the emissivity were 1.0?

Solution •

convert temperature to SI-unit: $[^{\circ}\text{C}] \rightarrow$

$$[\text{K}] T = (-10\text{ }^{\circ}\text{C}) + 273.15 = 263.15\text{ K}$$

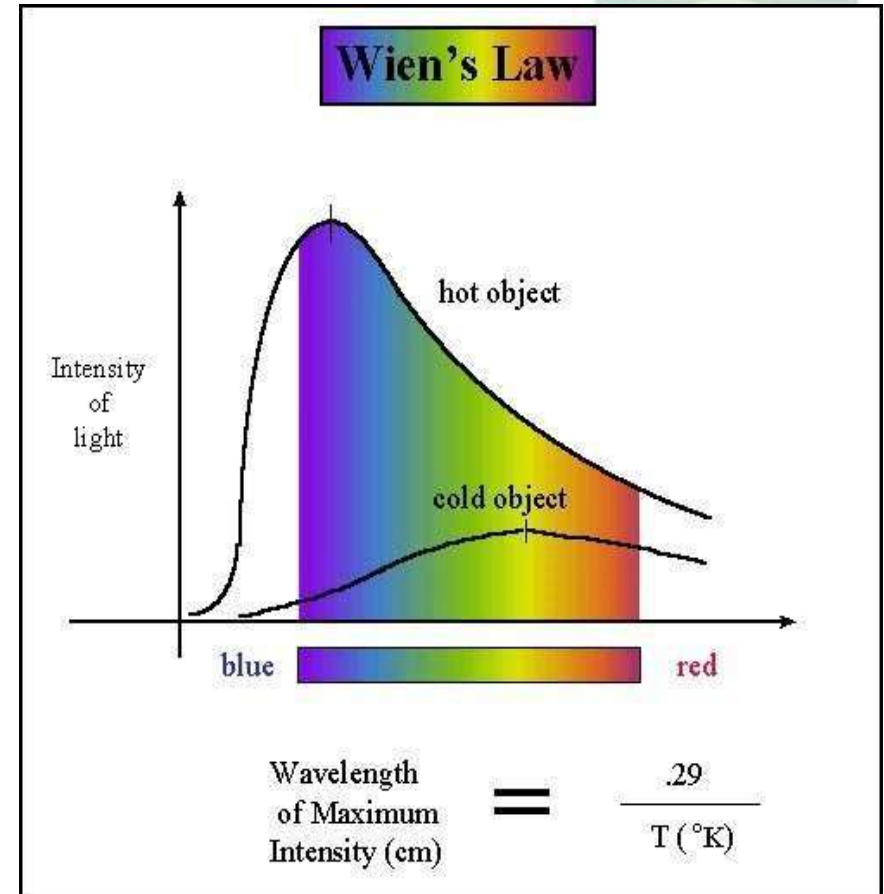
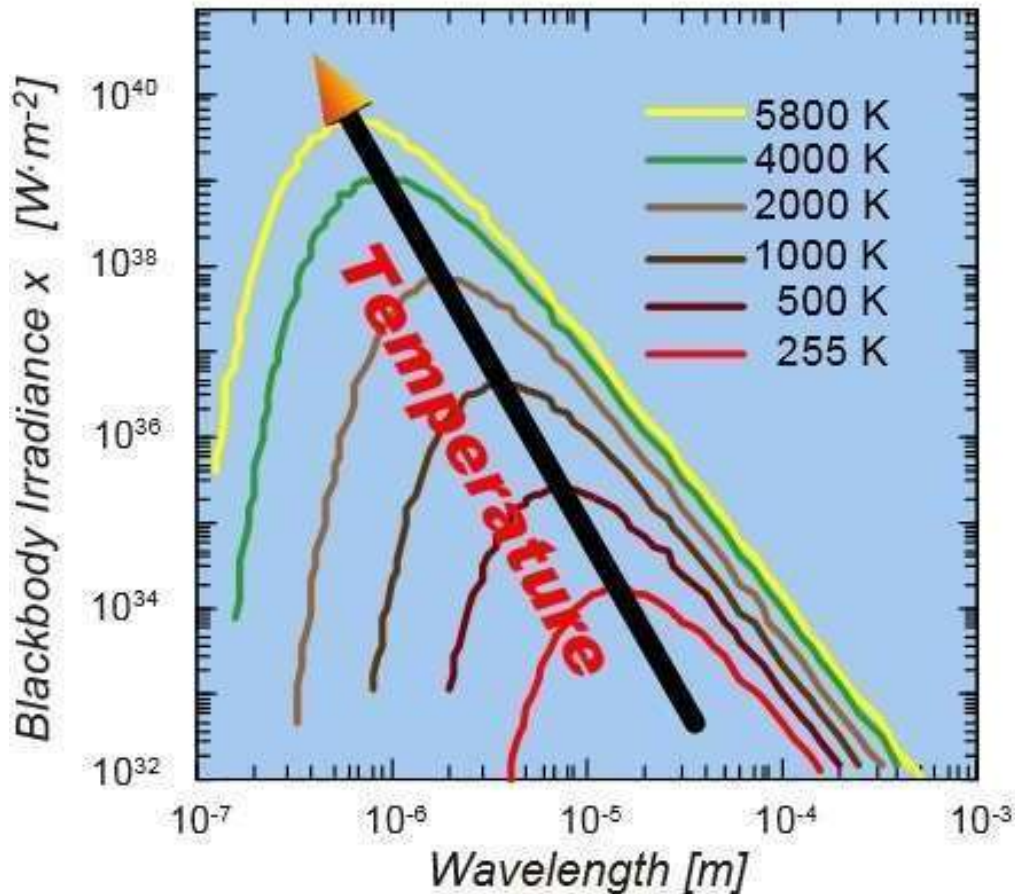
• use Stefan-Boltzmann law for $\epsilon = 1$ (black body): $F_{\text{cloud}} = \epsilon \cdot \sigma \cdot T^4 = 1 \times 5.67 \cdot 10^{-8} \times (263.15)^4$

$$= \boxed{271.9\text{ W m}^{-2}}$$

Check units: units okay – physics okay.

$$[\epsilon \cdot \sigma \cdot T^4] = [1] \times [\text{W m}^{-2} \text{K}^{-4}] \times [\text{K}^4] = [\text{W m}^{-2}]$$

3.3- LAWS OF RADIATION



3.3 -LAWS OF RADIATION



Example: 2

If a cloud bottom has a temperature of -10°C what is the wavelength of the peak energy emission? What part of the electromagnetic spectrum is this in?

Solution

- convert temperature to SI-unit: $[^{\circ}\text{C}] \rightarrow [\text{K}]$

$$T = (-10^{\circ}\text{C}) + 273.15 = 263.15 \text{ K}$$

- use Wien's law:

$$\lambda_{\text{max}} = a \cdot T^{-1} = 2898 \div 263.15 =$$

$$\boxed{11.0 \mu\text{m}}$$

Check units: *units okay – physics okay.* $[a \cdot T^{-1}]$

$$= [\mu\text{m} \cdot \text{K}] \times [\text{K}^{-1}] = [\mu\text{m}]$$



4- ATMOSPHERIC INFLUENCES ON RADIATION

Introduction

Global Shortwave Radiation Balance (overview)

- ~ 30 % of solar radiation is **reflected** by clouds, atmospheric gases and the surface
- ~ 25 % of solar radiation is **absorbed by the atmosphere** (clouds, atmospheric gases, aerosol)
- ~ 45 % of solar radiation is **absorbed by the surface** (oceans, land surface)

Influence of Clouds on Shortwave Radiation Balance

Clear conditions (no clouds):

- ~ 70 % of solar radiation is **absorbed by the surface** (55% direct, 15% diffuse sky radiation)
- **only ~ 13 %** of solar radiation is reflected

Cloudy conditions (overcast):

- ~ 25 % of solar radiation is absorbed by the surface (4% direct, 21% diffuse sky radiation)
- **51 %** of solar radiation is **reflected**