

Introduction of X-Ray

X-rays are a type of electromagnetic radiation.

- Other types of electromagnetic radiation are radio waves, microwaves, infrared, visible light, ultraviolet, and gamma rays.
- The types of radiation are distinguished by the amount of energy carried by the individual photons.
- All electromagnetic radiation consists of photons, which are individual packets of energy. For example, a household light bulb emits about 10^{21} photons of light (non-ionizing radiation) per second.
- The energy carried by individual photons, which is measured in electron volts (eV), is related to the frequency of the radiation.

X-rays ionize atoms.

- The energy required for ionization varies with the material (e.g., 34 eV in air, 25 eV in tissue) but is generally in the range of several eV.
- A 100 keV X-ray can potentially create thousands of ions.
- X-rays originate from atomic electrons and from free electrons decelerating in the vicinity of atoms (i.e., Bremsstrahlung).

X-Ray Production

- Radiation-producing devices produce X-rays by accelerating electrons through an electrical voltage potential and stopping them in a target.
- Many devices that use a high voltage and a source of electrons produce X-rays as an unwanted byproduct of device operation. These are called *incidental* X-rays.

- Most X-ray devices emit electrons from a cathode, accelerate them with a voltage, and allow them to hit an anode, which emits X-ray photons.
- These X-ray photons can be categorized as Bremsstrahlung or Characteristic.

Bremsstrahlung X-Rays

- When electrons hit the anode, they decelerate or brake and emit *Bremsstrahlung* (meaning *braking radiation* in German).
- Bremsstrahlung is produced most effectively when small charged particles interact with large atoms, such as when electrons hit a tungsten anode.
- However, Bremsstrahlung can be produced with any charged particles and any target. For example, at research laboratories, Bremsstrahlung has been produced by accelerating protons and allowing them to hit hydrogen.

Characteristic X-Ray

- When electrons change from one atomic orbit to another, *characteristic X-rays* are produced.
- The individual photon energies are characteristic of the type of atom and can be used to identify very small quantities of a particular element.
- For this reason, they are important in analytical X-ray applications at research laboratories.

X-Ray Regulations

The regulations on the handling of X rays are very similar to the regulations on standard radiation protection. The X-ray regulations in the European Union apply to those X-ray tubes and X-ray installations in which electrons are accelerated at least to 5 keV and in which they are limited to a maximum energy of 1 MeV. All installations in which electrons can be accelerated to energies beyond 1MeV are subject to the regulations of standard radiation protection.

Devices and installations that produce unwanted radiation, like old-fashioned TV screens, where electrons are accelerated up to energies of something like 20 keV, do not require a license if a dose rate of 1 $\mu\text{Sv/h}$ at a distance of 10 cm from the surface is not exceeded or if they are approved by the competent authority by way of a design approval.

The X-ray regulations, of course, mainly concern X-ray tubes used for X-ray diagnosis and X-ray therapy on humans. It is desirable to obtain the best X-ray image available for a particular radiation exposure. At the same time one should try to reduce the radiation exposure by improving the X-ray detection system and image reconstruction without affecting the image quality. The radiation dose of the patient has to be documented. If the patient wants a copy for patients of the documentation about the received X-ray doses, it has to be provided to the patient