



## Heat Transfer

Text book :- Heat Transfer. Tenth edition by J.P. Holman.

### CH.1 Introduction

Heat transfer :- Is the science that seeks to predict the energy transfer that may take place between material bodies as a result of a temperature difference.

#### □ Conduction Heat Transfer.

- \* When a temperature gradient exists in a body there is an energy transfer from the high-temperature region to the low-temperature region.
- \* We say that the energy is transferred by conduction and that the heat-transfer rate per unit area is proportional to the normal temperature gradient.

$$\frac{Q}{A} \propto \frac{\partial T}{\partial x} \Rightarrow \frac{Q}{A} = -K \frac{\partial T}{\partial x} \Rightarrow \boxed{Q = -KA \frac{\partial T}{\partial x}} \rightarrow \text{Fourier's Law}$$

\*  $Q$  :- heat-transfer rate. (W) or (J/s)

\*  $\frac{\partial T}{\partial x}$  :- Temp. gradient.  $\Rightarrow \begin{matrix} \Delta T \rightarrow \text{Temp. difference} \\ \Delta x \rightarrow \text{thickness} \end{matrix}$

\*  $K$  :- Thermal conductivity. (W/m.C) or (W/m.K)  
 منطبق! إيجاد قيم  $K$  من الجداول.

\*  $A$  :- H.T. area.

\*  $\Delta T$  :-  $T_2 - T_1$  -----  $T_1 > T_2$

\* أي أنه درجة الحرارة تنتقل من الأعلى  $T_1$  إلى الأقل  $T_2$ .



\* Note ∴  $K$  (for gas)  $<$   $K$  (for liquid, solid)

\* Note ∴  $K$  can be determined experimentally.

$$\therefore \boxed{q = -KA \frac{\Delta T}{\Delta x}}$$

**Example:** The front of slab of lead ( $k = 35 \text{ W/M.C}$ ) The front is kept @  $(100 \text{ C})$  and the back @  $(55 \text{ C})$  If the slab has an area of  $0.45 \text{ m}^2$  and  $3 \text{ cm}$  thickness.

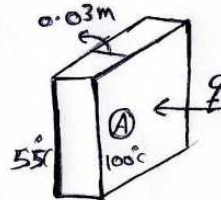
Determine the rate of heat transfer through this slab.

Solve ∴  $\Delta x = 0.03 \text{ m}$ ,  $\Delta T = (55 - 100)$ ,  $A = 0.45 \text{ m}^2$ ,  $K = 35 \text{ W/m.c}$

$$q = -KA \frac{\Delta T}{\Delta x}$$

$$q = -35 * 0.45 * \frac{(55 - 100)}{0.03}$$

$$q = 23625 \text{ W} = 23.625 \text{ Kw}$$



### EXAMPLE 1-1

One face of a copper plate  $3 \text{ cm}$  thick is maintained at  $400^\circ\text{C}$ , and the other face is maintained at  $100^\circ\text{C}$ . How much heat is transferred through the plate?

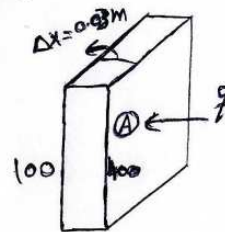
Solve ∴  $\Delta x = 0.03 \text{ m}$ ,  $\Delta T = (100 - 400)$

→  $K \rightarrow$  From table A.2  $K_{\text{copper}} = 370 \text{ W/m.c}$

$$q = -KA \frac{\Delta T}{\Delta x}$$

$$\frac{q}{A} = -K \frac{\Delta T}{\Delta x} = \frac{-370 * (100 - 400)}{0.03}$$

$$\frac{q}{A} = 3700000 \text{ W/m}^2$$





P. 1-1 If 3 kW is conducted through a section of insulating material 0.6 m<sup>2</sup> in cross section and 2.5 cm thick and the thermal conductivity may be taken as 0.2 W/m · °C, compute the temperature difference across the material.

Soly :- Find  $\Delta T$  ?  $q = 3 \text{ kW} = 3000 \text{ W}$ ,  $\Delta x = 2.5 \text{ cm} = 0.025 \text{ m}$

في هذا السؤال ليس قانون بيرنولي بل قانون انتقال الحرارة بالتوصيل  
 لأن انتقال الحرارة بالتوصيل عبر مادة صلبة.

$$q = -k A \frac{\Delta T}{\Delta x}$$

$$q = k A \frac{\Delta T}{\Delta x} \Rightarrow \Delta T = \frac{q \cdot \Delta x}{k A} = \frac{3000 \times 0.025}{0.2 \times 0.6} = 625^\circ \text{C}$$

P. 1-2 A temperature difference of 85°C is impressed across a fiberglass layer of 13 cm thickness. The thermal conductivity of the fiberglass is 0.035 W/m · °C. Compute the heat transferred through the material per hour per unit area.

Soly :- Find  $\frac{q}{A}$  in J/h.m<sup>2</sup>.  $\Delta T = 85^\circ \text{C}$ ,  $\Delta x = 13 \text{ cm} = 0.13 \text{ m}$ ,  $k = 0.035 \frac{\text{W}}{\text{m} \cdot ^\circ \text{C}}$

انظر للملاحظة في السؤال السابق.

$$\frac{q}{A} = k \frac{\Delta T}{\Delta x} = \frac{0.035 \times 85}{0.13} = 22.885 \frac{\text{W}}{\text{m}^2} = 22.885 \frac{\text{J}}{\text{s} \cdot \text{m}^2}$$

$$\frac{q}{A} = 22.885 \frac{\text{J}}{\text{s} \cdot \text{m}^2} \times \frac{60 \text{ s}}{1 \text{ min}} \times \frac{60 \text{ min}}{1 \text{ h}} \Rightarrow \frac{q}{A} = 82386 \text{ J/h} \cdot \text{m}^2$$

P. 1-4 The temperatures on the faces of a plane wall 15 cm thick are 375 and 85°C. The wall is constructed of a special glass with the following properties:  $k = 0.78 \text{ W/m} \cdot ^\circ \text{C}$ ,  $\rho = 2700 \text{ kg/m}^3$ ,  $c_p = 0.84 \text{ kJ/kg} \cdot ^\circ \text{C}$ . What is the heat flow through the wall at steady-state conditions?

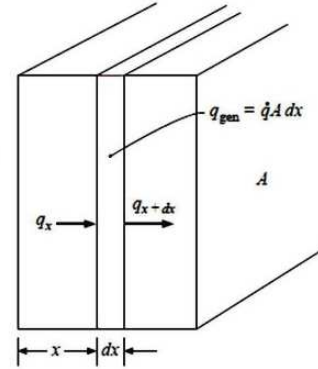
Soly :- Find  $\frac{q}{A}$  ?  $\Delta x = 15 \text{ cm} = 0.15 \text{ m}$ ,  $k = 0.78$

$$\frac{q}{A} = -k \frac{\Delta T}{\Delta x} = -0.78 \times \frac{(85 - 375)}{0.15} \Rightarrow \frac{q}{A} = 1508 \text{ W/m}^2$$

\* General Heat conduction equation

$$\begin{aligned}
 - dV &= A dx \\
 - dm &= \rho dV \\
 \Rightarrow dm &= \rho A dx \\
 \therefore m &= \rho \Delta V \\
 m &= \rho A dx
 \end{aligned}$$

**Figure 1-2 |**  
 Elemental volume for one-dimensional heat-conduction analysis.



\* Energy Balance For the Element

- 1 Heat conducted through left face + 2 Heat generated within element =  
 3 Heat conducted from right face + 4 Heat stored within the element [Internal Energy]

$$\Rightarrow 1 \quad \dot{q}_x = -KA \frac{\partial T}{\partial x}$$

$$2 \quad \dot{q}_{gen} = \dot{q} A dx$$

- rate of heat generated per unit volume
- Electric current passes through a resistance
- Exothermic reaction
- Nuclear reaction.
- (W/m<sup>3</sup>)

$$\begin{aligned}
 3 \quad \dot{q}_{x+dx} &= \dot{q}_x + \frac{\partial}{\partial x} (\dot{q}_x) dx \\
 &= -KA \frac{\partial T}{\partial x} + \frac{\partial}{\partial x} \left( -KA \frac{\partial T}{\partial x} \right) dx
 \end{aligned}$$

$$\begin{aligned}
 4 \quad \dot{q}_{stored} &= m \cdot c \cdot \frac{\partial T}{\partial t} \\
 &= \rho A dx c \frac{\partial T}{\partial t}
 \end{aligned}$$

$c$  = specific heat From  
 $\rho$  = Density  
 $t$  = Time



$$\boxed{1} + \boxed{2} = \boxed{3} + \boxed{4}$$

$$\frac{\partial}{\partial x} \left[ k \frac{\partial T}{\partial x} \right] + \dot{q}'' = \rho c \frac{\partial T}{\partial t}$$

$$\frac{\partial^2 T}{\partial x^2} + \frac{\dot{q}''}{k} = \frac{1}{\alpha} \frac{\partial T}{\partial t}$$

General one-dimensional conduction Heat transfer.

\*  $\alpha$  Thermal Diffusivity  $m^2/s$

$$\alpha = \frac{k}{\rho c}$$

\* For three-dimensional H.T. conduction:

$$\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} + \frac{\dot{q}''}{k} = \frac{1}{\alpha} \frac{\partial T}{\partial t}$$

**Example:** A 0.5m thickness wall made from a material

(  $c_p = 5000 \text{ J/kg.k}$  ,  $\rho = 1600 \text{ kg/m}^3$  ,  $k = 50 \text{ w/m.k}$  ) ,  
 temp distribution is given by  $T(x) = a + bx + cx^2$   $a = 700$   
 $C^\circ$  ,  $b = -200 \text{ C}^\circ/\text{m}$  ,  $c = -20 \text{ C}^\circ/\text{m}^2$  ,  $A = 10\text{m}^2$  ,  $\dot{q}'' =$   
 $2000 \text{ w/m}^3$  Determine :

1- Rate of H.T entering and leaving the wall .

2- Rate of storing energy within the wall.

3- Rate of change of temperature @  $x = 0.125\text{m}$

$x = 0.25 \text{ m}$



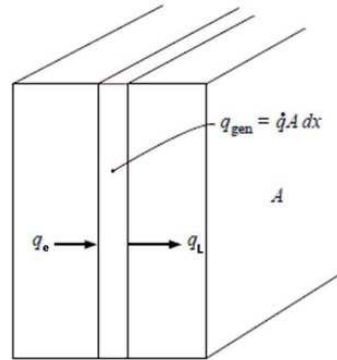
Soly

اولا قبل ما نحل المسألة استقر الحرارة مره

$$T(x) = 700 - 200x - 20x^2$$

$$\frac{\partial T}{\partial x} = -200 - 40x$$

$$\frac{\partial^2 T}{\partial x^2} = -40 \quad , \quad \frac{\partial^3 T}{\partial x^3} = 0$$



$$\textcircled{1} \quad \dot{q}_x \text{ (entering)} = -kA \left. \frac{\partial T}{\partial x} \right|_{x=0} = -50 \times 10 \times [-200 - 40(0)]$$

$$\boxed{\dot{q}_x = 100 \text{ kW}}$$

$$-\dot{q}_{x+dx} \text{ (leaving)} = -kA \left. \frac{\partial T}{\partial x} \right|_{x=0.5} + \frac{\partial}{\partial x} \left( -kA \frac{\partial T}{\partial x} \right) dx$$

$$\dot{q}_{x+dx} = -50 \times 10 \times [-200 - 40(0.5)] \Rightarrow \boxed{\dot{q}_{x+dx} = 110 \text{ kW}}$$

$$\textcircled{2} \quad \dot{q}_x + \dot{q}_g = \dot{q}_{x+dx} + \dot{q}_{\text{stored}}$$

$$\dot{q}_{\text{st}} = \dot{q}_x + \dot{q}_g - \dot{q}_{x+dx}$$

$$\dot{q}_{\text{st}} = 100 + 10 - 110$$

$$\boxed{\dot{q}_{\text{st}} = 0}$$

$$\dot{q}_g = \dot{q} \Delta V = \dot{q} \Delta x A$$

$$\dot{q}_g = 2000 \times 0.5 \times 10$$

$$\boxed{\dot{q}_g = 10 \text{ kW}}$$

$$\textcircled{3} \quad \frac{\partial^2 T}{\partial x^2} + \frac{\dot{q}}{k} = \frac{1}{\alpha} \frac{\partial T}{\partial t}$$

Find.  $\left. \frac{\partial T}{\partial t} \right|_{x=0.125}$   
 $x=0.25$

$$\alpha = \frac{k}{\rho c} = \frac{50}{1600 \times 5000}$$

$$\boxed{\alpha = 6.25 \times 10^{-6} \text{ m}^2/\text{s}}$$

استنتج  $\frac{\partial^2 T}{\partial x^2} = -40$  @ any value of  $x$

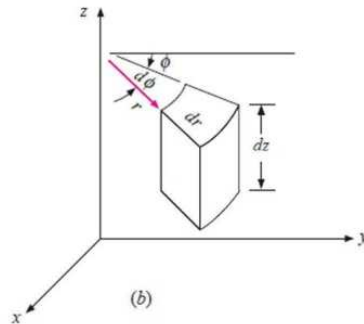
ثم تعويض مباشر المعادلة  $\frac{\partial T}{\partial r} = \left[ -40 + \frac{20000}{50} \right] * 6.25 * 10^{-6}$

$\frac{\partial T}{\partial r} = 0$

\* Steady state conduction.  
 \* درجة الحرارة لا تتغير مع الزمن.

\* Cylindrical coordinates. (3-dimension) Fig 1-3(b)

$$\frac{\partial^2 T}{\partial r^2} + \frac{1}{r} \frac{\partial T}{\partial r} + \frac{1}{r^2} \frac{\partial^2 T}{\partial \theta^2} + \frac{\partial^2 T}{\partial z^2} + \frac{q'''}{k} = \frac{1}{\alpha} \frac{\partial T}{\partial t}$$



\* Spherical coordinates (3-dimension) Fig 1-3(c)

$$\frac{1}{r} \frac{\partial^2}{\partial r^2} (rT) + \frac{1}{r^2 \sin \theta} \frac{\partial}{\partial \theta} \left[ \sin \theta \frac{\partial T}{\partial \theta} \right] + \frac{1}{r^2 \sin^2 \theta} \frac{\partial^2 T}{\partial \phi^2} + \frac{q'''}{k} = \frac{1}{\alpha} \frac{\partial T}{\partial t}$$

