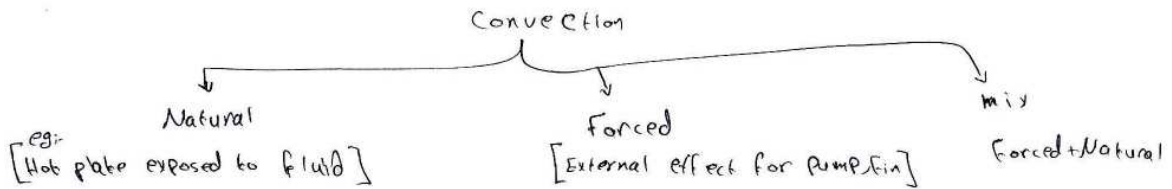




2 Convection Heat Transfer.

It is well known that a hot plate of metal will cool faster when placed in front of a fan than when exposed to still air. We say that the heat is convected away, and we call the process *convection heat transfer*.



سائل أو الغاز
 fluid ← gas liquid

* Newton cooling law.

$$* \dot{Q} = hA(T_w - T_\infty)$$

- \dot{Q} is rate of H.T

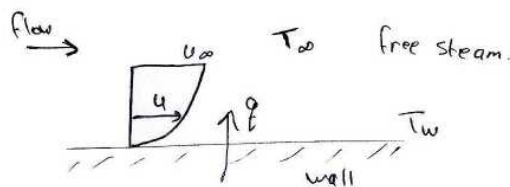
- h is convection H.T coefficient $\cdot (W/m^2 \cdot C)$ OR $(W/m^2 \cdot K)$

- A is H.T area

- T_w is wall temp

- T_∞ is Fluid temp.

* Fig 1-7 :- convection H.T. from a plate.



* Note :-

| | | |
|---|-------------------|-------------------|
| h | Free convection | gas : 2 - 25 |
| | | liq : 50 - 1000 |
| h | Forced convection | gas : 25 - 250 |
| | | liq : 100 - 20000 |

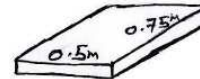
EXAMPLE 1-2

Air at 20°C blows over a hot plate 50 by 75 cm maintained at 250°C. The convection heat-transfer coefficient is 25 W/m²·°C. Calculate the heat transfer.

Soly ∴ Find q ∴ $A = 0.5 \times 0.75$, $h = 25$, $T_w = 250^\circ\text{C}$, $T_\infty = 20^\circ\text{C}$

$$q = hA(T_w - T_\infty) \Rightarrow q = 25 \times 0.5 \times 0.75 (250 - 20)$$

$$q = 2156.25 \text{ W}$$



EXAMPLE 1-4

An electric current is passed through a wire 1 mm in diameter and 10 cm long. The wire is submerged in liquid water at atmospheric pressure, and the current is increased until the water boils. For this situation $h = 5000 \text{ W/m}^2 \cdot ^\circ\text{C}$, and the water temperature will be 100°C. How much electric power must be supplied to the wire to maintain the wire surface at 114°C?

Soly ∴ Find q , $D = 1 \text{ mm}$, $L = 10 \text{ cm}$, $h = 5000 \frac{\text{W}}{\text{m}^2 \cdot ^\circ\text{C}}$, $T_\infty = 100^\circ\text{C}$
 $T_w = 114^\circ\text{C}$.

$$\Rightarrow q = hA(T_w - T_\infty)$$

$$- A = \pi DL = \pi \times 0.001 \times 0.1 = 3.142 \times 10^{-4} \text{ m}^2$$

$$\Rightarrow q = 5000 \times 3.142 \times 10^{-4} \times (114 - 100)$$

$$q = 21.994 \text{ W} \equiv \text{Electric power must be applied}$$



* Convection in a channel

$$q = \dot{m} (i_e - i_i) = \dot{m} \Delta i$$

$$\Delta i = c_p \Delta T = c_p (T_e - T_i)$$

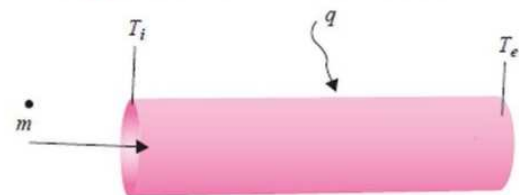
i → enthalpy.

\dot{m} → fluid mass flow rate kg/s

$$q = \dot{m} c_p (T_e - T_i) = hA (T_w - T_\infty)$$

c_p ∴ specific heat in J/kg·°C

Figure 1-8 | Convection in a channel.





* T_e, T_i, T_∞ are called bulk Temp.

* $\dot{m} = \rho U_{mean} A_c$

$\rho \rightarrow$ density kg/m^3

$U_{mean} \rightarrow$ velocity m/s

$A_c \rightarrow$ cross sectional area \rightarrow For circular Tube
 $A_c = \frac{\pi d^2}{4}$

* A is Surface area for convection in this case is,

$q = hA(T_w - T_\infty)$

$A = \pi dL$

$L \rightarrow$ length of tube.



P. 1-16

Water flows at the rate of 0.5 kg/s in a 2.5-cm-diameter tube having a length of 3 m. A constant heat flux is imposed at the tube wall so that the tube wall temperature is 40°C higher than the water temperature. Calculate the heat transfer and estimate the temperature rise in the water. The water is pressurized so that boiling cannot occur.

Soly :- Find ① q ② $(T_e - T_i)$

$T_w - T_\infty = 40^\circ\text{C}$ أنته

$h \rightarrow$ from Table 1-3. $\Rightarrow h = 3500 \text{ W/m}^2\text{K}$

① $q = hA(T_w - T_\infty) = h\pi dL(T_w - T_\infty)$

$q = 3500 \times \pi \times 0.025 \times 3 \times 40 \Rightarrow q = 32.9867 \text{ kW}$

② $q = \dot{m} c_p (T_e - T_i)$

given $c_{p \text{ water}} = 4186 \text{ J/kg}\cdot\text{C}$

$T_e - T_i = \frac{q}{\dot{m} c_p} = \frac{32986.7}{0.5 \times 4186} = 15.76^\circ\text{C}$



P. 1-23 A flat wall is exposed to an environmental temperature of 38°C . The wall is covered with a layer of insulation 2.5 cm thick whose thermal conductivity is $1.4\text{ W/m}\cdot^{\circ}\text{C}$, and the temperature of the wall on the inside of the insulation is 315°C . The wall loses heat to the environment by convection. Compute the value of the convection heat-transfer coefficient that must be maintained on the outer surface of the insulation to ensure that the outer-surface temperature does not exceed 41°C .

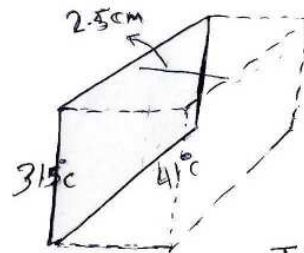
Soly: Find h

الحرارة المنقولة بالمواد = الحرارة المنقولة بالتيار
 conduction = convection

$$-kA \frac{\Delta T}{\Delta x} = hA(T_w - T_{\infty})$$

$$-1.4 \times \frac{(41 - 315)}{0.025} = h(41 - 38)$$

$$h = 5114.66\text{ W/m}^2\cdot^{\circ}\text{C}$$



$$T_{\infty} = 38^{\circ}\text{C}$$

$$K = 1.4\text{ W/m}\cdot^{\circ}\text{C}$$

P. 1-26 How does the free-convection heat transfer from a vertical plate compare with pure conduction through a vertical layer of air having a thickness of 2.5 cm and a temperature difference the same at $T_w - T_{\infty}$? Use information from Table 1-3.

Soly: -1 Free convection from vertical plate

$$q = hA(T_w - T_{\infty}) = hA\Delta T$$

⇒ From Table 1-3

المسألة: [free convection $\Delta T = 30^{\circ}\text{C}$
 vertical plate 0.3 m high in air] → $h = 4.5\text{ W/m}^2\cdot^{\circ}\text{C}$

$$q = 4.5 \times (0.3)^2 \times 30 \Rightarrow q = 12.15\text{ W}$$

-2 conduction through a vertical layer of air $\Delta x = 2.5\text{ cm}$

$$\Delta T = (T_w - T_{\infty}) = 30^{\circ}\text{C}$$

⇒ From table A-5

$$q = kA \frac{\Delta T}{\Delta x} = 0.027 \times (0.3)^2 \times \frac{30}{0.025}$$

$$q = 2.916\text{ W}$$

تحويل درجة الحرارة إلى كلفين

$$T(K) = 273.15 + 30 = 303.15\text{ Kelvin}$$

$$T(K) \Leftrightarrow K = 0.027\text{ W/m}\cdot^{\circ}\text{C}$$



3] Radiation Heat transfer

- opposite to conduction and convection.
- Radiation may achieved in vacuum.
- It depends on electro magnetic waves.

* Thermal Ideal radiation.

Black Body ☺☺

$$q = \sigma A T^4 \Rightarrow \text{Stefan - Boltzmann law}$$

- q ☺☺ rate of heat transfer from a Black Body.
- σ ☺☺ stefan-Boltzmann constant
 $\sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \cdot \text{K}^4$
- A ☺☺ heat transfer area.
- T ☺☺ Absolute temp (Kelvin)

P. 1-20 Calculate the energy emitted by a blackbody at 1000°C .

Soly ☺☺ $q = \sigma A T^4 \Rightarrow \frac{q}{A} = \sigma T^4 = 5.67 \times 10^{-8} \times (1000 + 273.15)^4$

$$\frac{q}{A} = 1.4897 \times 10^5 \text{ W/m}^2$$

P.1-21 If the radiant flux from the sun is 1350 W/m^2 , what would be its equivalent blackbody temperature?

Soly ☺☺ Find T ? , $q/A = 54 \times 10^6 \text{ W/m}^2$

$$T = \sqrt[4]{\frac{q}{\sigma A}} \Rightarrow T = 5555.23 \text{ K}$$



* Heat Exchange Between Two Black Surface so.

$$Q = \sigma A (T_1^4 - T_2^4) \quad T_1, T_2 \rightarrow \text{in Kelvin}$$

P. 1-13 Two very large parallel planes having surface conditions that very nearly approximate those of a blackbody are maintained at 1100 and 425°C, respectively. Calculate the heat transfer by radiation between the planes per unit time and per unit surface area.

Soly so: $T_1 = 1100^\circ\text{C} = 1373 \text{ K}$
 $T_2 = 425^\circ\text{C} = 698 \text{ K}$ Find Q/A (W/m²) . ?

$$\frac{Q}{A} = \sigma (T_1^4 - T_2^4) \Rightarrow \boxed{\frac{Q}{A} = 188 \text{ kW/m}^2}$$

* Radiation in an Enclosure.

$$Q = \sigma A \epsilon (T_1^4 - T_2^4) \quad \text{Black Body } \epsilon = 1$$

$$\epsilon \rightarrow \text{Emissivity, } 0 < \epsilon < 1 \rightarrow \text{from App. A.}$$

P. 1-19 A small radiant heater has metal strips 6 mm wide with a total length of 3 m. The surface emissivity of the strips is 0.85. To what temperature must the strips be heated if they are to dissipate 2000 W of heat to a room at 25°C?

Soly so: width = 0.006 m, length = 3 m, $\epsilon = 0.85$, $Q = 2000 \text{ W}$, $T_2 = 298 \text{ K}$

$$Q = \sigma A \epsilon (T_1^4 - T_2^4) \Rightarrow 2000 = 5.67 \times 10^{-8} \times 0.006 \times 3 \times 0.85 \times (T_1^4 - 298^4)$$

$$\boxed{T_1 = 1233.7 \text{ K}}$$

EXAMPLE 1.3

Assuming that the plate in Example 1-2 is made of carbon steel (1%) 2 cm thick and that 300 W is lost from the plate surface by radiation, calculate the inside plate temperature.

Find ΔT ?

Solve

$$q_{\text{rad}} = 300 \text{ W}$$

$$q_{\text{conv}} = 2156 \text{ W}$$

From Table \Rightarrow $k_{\text{carbon}} = 43 \text{ W/m}\cdot\text{c}$

$$q_{\text{cond}} = q_{\text{conv}} + q_{\text{rad}}$$

$$-kA \frac{\Delta T}{\Delta x} = 2156 + 300$$

$$\Delta T = \frac{-2456 * 0.02}{0.5 * 0.75 * 43}$$

$$T_2 - T_1 = -3.05$$

$$T_1 = T_2 + 3.05 \Rightarrow T_1 = 250 + 3.05$$

$T_1 = 253.05^\circ\text{C}$

Figure 1-9 | Combination of conduction, convection, and radiation heat transfer.

